

Mineral geochemistry for precious and base metal exploration

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COESSING 2018
University of Ghana

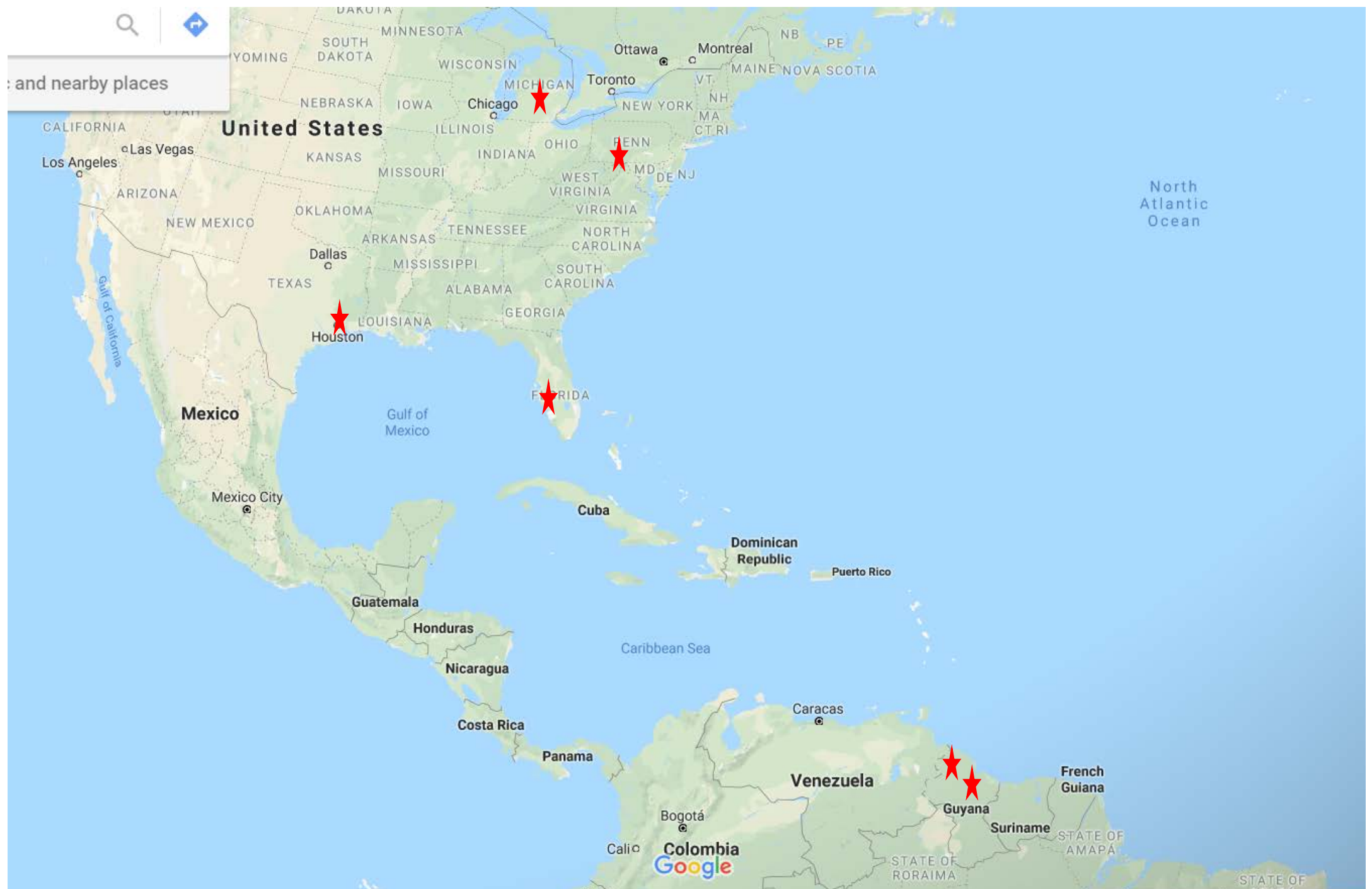
Who am I?/ Where I've lived?

- Matthew's Ridge, Guyana (home)
- Georgetown, Guyana
- Gettysburg, PA
- Tampa, FL
- Ann Arbor, MI
- Houston, TX





and nearby places



How do we feel about mining?

- Why do we mine?
- What do we mine?

ELEMENTS OF A SMARTPHONE

ELEMENTS COLOUR KEY: ● ALKALI METAL ● ALKALINE EARTH METAL ● TRANSITION METAL ● GROUP 13 ● GROUP 14 ● GROUP 15 ● GROUP 16 ● HALOGEN ● LANTHANIDE

SCREEN



Indium tin oxide is a mixture of indium oxide and tin oxide, used in a transparent film in the screen that conducts electricity. This allows the screen to function as a touch screen.



The glass used on the majority of smartphones is an aluminosilicate glass, composed of a mix of alumina (Al_2O_3) and silica (SiO_2). This glass also contains potassium ions, which help to strengthen it.



A variety of Rare Earth Element compounds are used in small quantities to produce the colours in the smartphone's screen. Some compounds are also used to reduce UV light penetration into the phone.

ELECTRONICS



Copper is used for wiring in the phone, whilst copper, gold and silver are the major metals from which microelectrical components are fashioned. Tantalum is the major component of micro-capacitors.



Nickel is used in the microphone as well as for other electrical connections. Alloys including the elements praseodymium, gadolinium and neodymium are used in the magnets in the speaker and microphone. Neodymium, terbium and dysprosium are used in the vibration unit.



Pure silicon is used to manufacture the chip in the phone. It is oxidised to produce non-conducting regions, then other elements are added in order to allow the chip to conduct electricity.



Tin & lead are used to solder electronics in the phone. Newer lead-free solders use a mix of tin, copper and silver.

BATTERY



The majority of phones use lithium ion batteries, which are composed of lithium cobalt oxide as a positive electrode and graphite (carbon) as the negative electrode. Some batteries use other metals, such as manganese, in place of cobalt. The battery's casing is made of aluminium.

CASING



Magnesium compounds are alloyed to make some phone cases, whilst many are made of plastics. Plastics will also include flame retardant compounds, some of which contain bromine, whilst nickel can be included to reduce electromagnetic interference.



Objectives

- To provide an introduction to economic geology and natural resources
- Introduction to using geochemistry for exploration
- To (hopefully) change the way you think about (geo)chemistry and natural resources

Geology 101

- Rocks are our friends.
- Atoms – elements – minerals – rocks



What is Economic Geology?



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The Free Encyclopedia

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Economic geology

From Wikipedia, the free encyclopedia

"Economic Geology" redirects here. For the journal, see [Economic Geology \(journal\)](#).

Economic geology is concerned with earth materials that can be used for economic and/or industrial purposes. These materials include precious and base metals, nonmetallic minerals, construction-grade stone, petroleum minerals, coal, and water. Economic geology is a subdiscipline of the geosciences; according to Lindgren (1933) it is "the application of geology". Today, we might call it the scientific study of the Earth's sources of mineral raw materials and the practical application of the acquired knowledge.^[1] The term commonly refers to metallic mineral deposits and mineral resources. The techniques employed by other earth science disciplines (such as [geochemistry](#), [mineralogy](#), [geophysics](#), [petrology](#) and [structural geology](#)) might all be used to understand, describe, and exploit an ore deposit.

Economic geology is studied and practiced by geologists. Economic geology may be of interest to other professions such as engineers, environmental scientists, and conservationists because of the far-reaching impact that extractive industries have on society, the economy, and the environment.

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- [Purpose of studies](#)
- [Mineral resources](#)
- [Ore geology](#)
- [Coal and petroleum geology](#)
- [See also](#)
- [References](#)

PERIODIC TABLE *of the* ELEMENTS



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VII A 18
He
Helium 2
4.00

IA 1 H Hydrogen 1 1.01	IIA 2 Li Lithium 3 6.94	Be Beryllium 4 9.01	Mg Magnesium 12 24.31	Na Sodium 11 22.99	K Potassium 19 39.10	Ca Calcium 20 40.08	Rb Rubidium 37 85.47	Cs Caesium 55 132.91	Fr Francium 87 (223)												
Sc Scandium 21 44.96	Ti Titanium 22 47.88	V Vanadium 23 50.94	Cr Chromium 24 52.00	Mn Manganese 25 54.94	Fe Iron 26 55.85	Co Cobalt 27 58.93	Ni Nickel 28 58.69	Cu Copper 29 63.55	Zn Zinc 30 65.39	Ga Gallium 31 69.72	Ge Germanium 32 72.61	As Arsenic 33 74.92	Se Selenium 34 78.96	Br Bromine 35 79.90	Kr Krypton 36 83.80	Xe Xenon 54 131.29	Rn Radon 86 (222)				
Y Yttrium 39 88.91	Zr Zirconium 40 91.22	Nb Niobium 41 92.91	Mo Molybdenum 42 95.94	Tc Technetium 43 (98)	Ru Ruthenium 44 101.07	Rh Rhodium 45 102.91	Pd Palladium 46 106.42	Ag Silver 47 107.87	Cd Cadmium 48 112.41	In Indium 49 114.82	Sn Tin 50 118.71	Sb Antimony 51 121.76	Te Tellurium 52 127.60	I Iodine 53 126.90	Xe Xenon 54 131.29	Rn Radon 86 (222)					
Ba Barium 56 137.33	Hf Hafnium 72 178.49	Ta Tantalum 73 180.95	W Tungsten 74 183.85	Re Rhenium 75 186.21	Os Osmium 76 190.23	Ir Iridium 77 192.22	Pt Platinum 78 195.08	Au Gold 79 196.97	Hg Mercury 80 200.59	Tl Thallium 81 204.38	Pb Lead 82 207.20	Bi Bismuth 83 208.98	Po Polonium 84 (209)	At Astatine 85 (210)	Rn Radon 86 (222)						
Ra Radium 88 (226)	Rf Rutherfordium 104 (261)	Db Dubnium 105 (262)	Sg Seaborgium 106 (263)	Bh Bohrium 107 (262)	Hs Hassium 108 (265)	Mt Meitnerium 109 (266)	La Lanthanum 57 138.91	Ce Cerium 58 140.12	Pr Praseodymium 59 140.90	Nd Neodymium 60 144.24	Pm Promethium 61 (145)	Sm Samarium 62 150.36	Eu Europium 63 151.96	Gd Gadolinium 64 157.25	Tb Terbium 65 158.93	Dy Dysprosium 66 162.50	Ho Holmium 67 164.93	Er Erbium 68 167.26	Tm Thulium 69 168.93	Yb Ytterbium 70 173.05	Lu Lutetium 71 174.96
Ac Actinium 89 227.03	Th Thorium 90 232.04	Pa Protactinium 91 231.04	U Uranium 92 238.03	Np Neptunium 93 237.05	Pu Plutonium 94 244.06	Am Americium 95 243.06	Cm Curium 96 247.07	Bk Berkelium 97 247.07	Cf Californium 98 251.08	Es Einsteinium 99 252.08	Fm Fermium 100 257.10	Md Mendelevium 101 258.10	No Nobelium 102 259.10	Lr Lawrencium 103 260.10							

ALKAALI METALS
ALKAALI EARTH METALS
TRANSITION METALS
OTHER METALS
NON-METALS
HALOGENS
NOBLE GASSES
DIATOME METALS

At room temperature the element is:
 ● Gas
 ● Liquid
 ● Natural solid
 ● Man-made solid (synthetic)

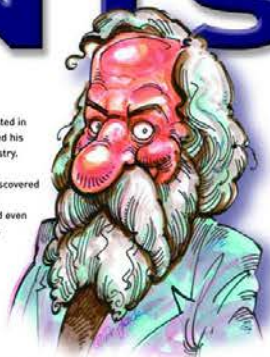
Symbol
Element name
Atomic number
Atomic mass

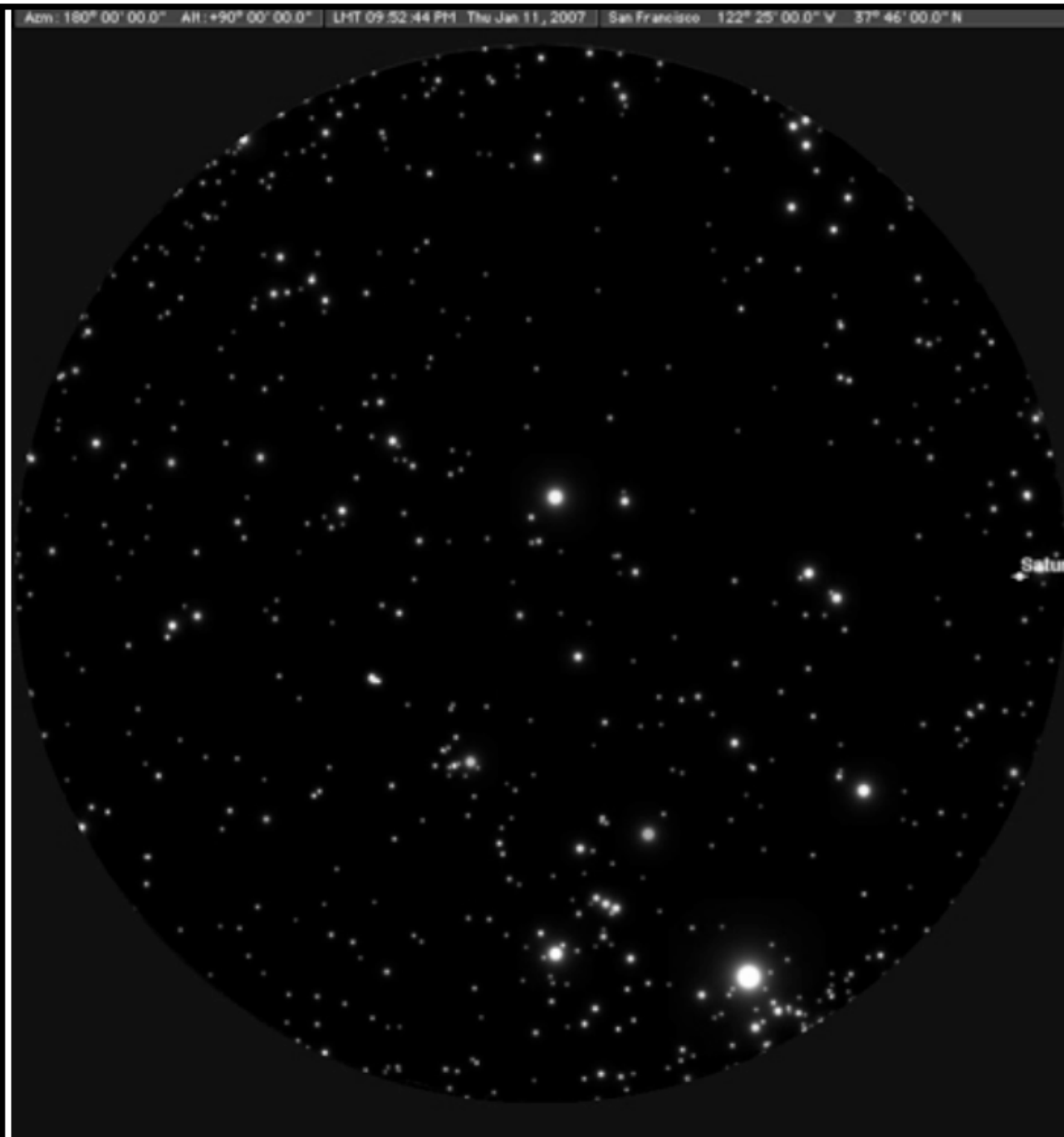
DMITRI MENDELEYEV (1834 - 1907)

The Russian chemist, Dmitri Mendeleev, was the first to observe that if elements were listed in order of atomic mass, they showed regular (periodical) repeating properties. He formulated his discovery in a periodic table of elements, now regarded as the backbone of modern chemistry.

The crowning achievement of Mendeleev's periodic table lay in his prophecy of then, undiscovered elements. In 1869, the year he published his periodic classification, the elements gallium, germanium and scandium were unknown. Mendeleev left spaces for them in his table and even predicted their atomic masses and other chemical properties. Six years later, gallium was discovered and his predictions were found to be accurate. Other discoveries followed and their chemical behaviour matched that predicted by Mendeleev.

This remarkable man, the youngest in a family of 17 children, has left the scientific community with a classification system so powerful that it became the cornerstone in chemistry teaching and the prediction of new elements ever since. In 1955, element 101 was named after him: Md, Mendeleevium.





When you look
at stars in a
clear night sky,
what differences
can you see?

Periodic Groups and Natural Systems

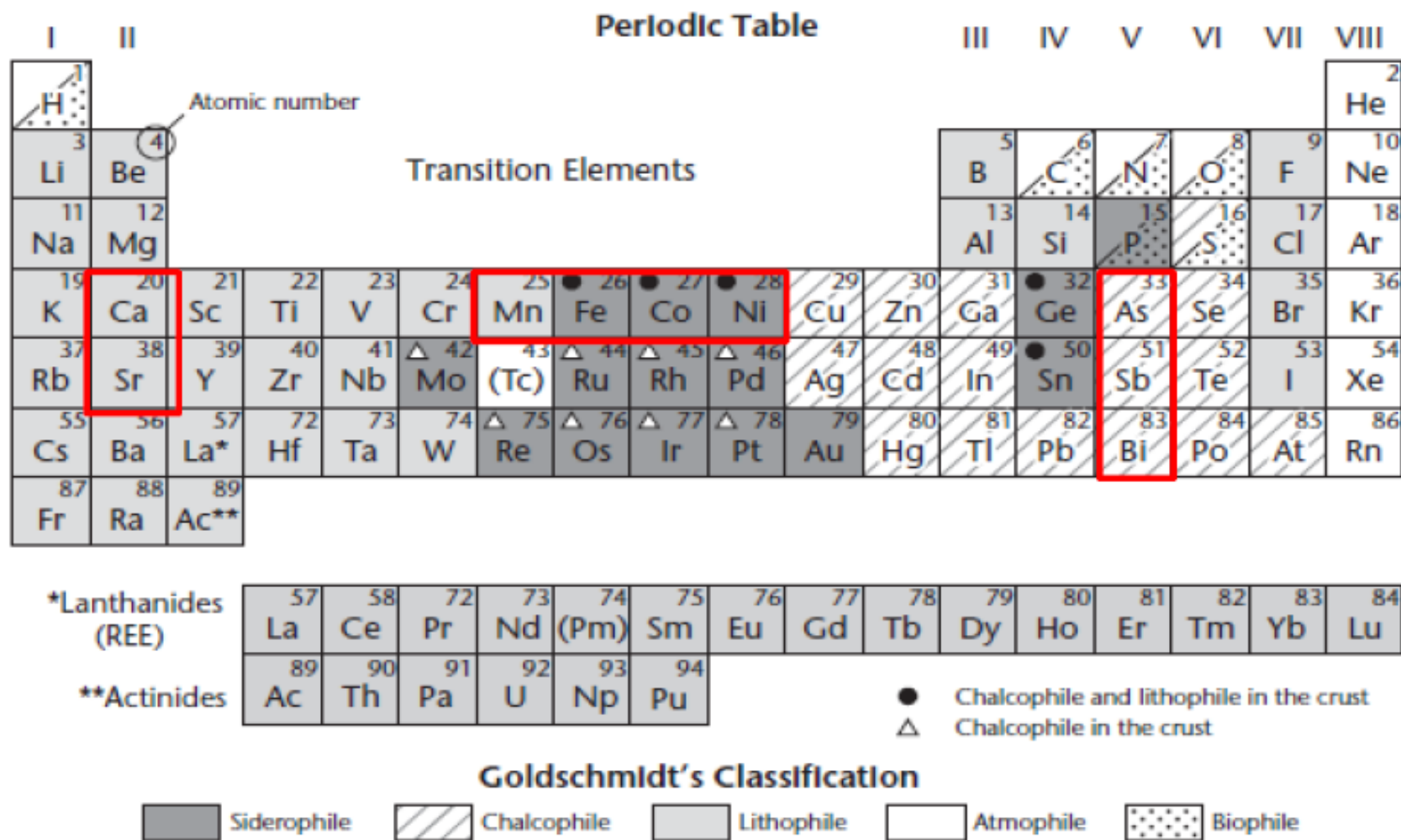


Figure 5.1: Periodic Table with superimposed Goldschmidt Classification.

Copper (Cu)

- Porphyry copper deposits
 - Stratiform Cu deposits
 - Iron oxide copper gold deposits
- deposits

Chile, Peru, DRC, Zambia, USA



chalcopyrite



Iron (Fe): magnetite (Fe_3O_4) & hematite (Fe_2O_3)

- Manufacture of steel
- Banded iron formations
- Iron oxide copper gold deposits
- Iron oxide apatite deposits

- China, Australia, Brasil



Gold (Au)

- Jewelry
- Finances
- Electronics
- Dentistry and medicine

- Orogenic gold deposits
- Porphyry deposits
- Placer deposits

- China, South Africa, Ghana, Russia, USA, Canada, Australia



Aluminium (Al) [Al₂O₃]

- Bauxite
- Australia, China, Brasil



Phosphorus (P) – $\text{Ca}_5(\text{PO}_4)_3(\text{F}, \text{Cl}, \text{OH})$

- Agriculture: Fertilizers and supplements for animals
- Construction
- Pharmaceuticals
- Water treatment

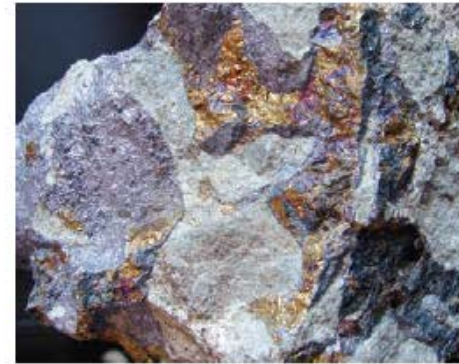
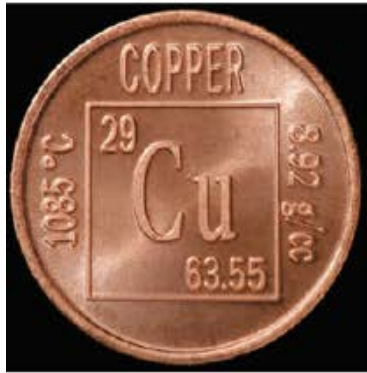
- Phosphorite
- Apatite

- USA, China, Morocco



Economic Geology: From Background to Ore

<i>Element</i>	<i>Oxidation state</i>	<i>Deposit type</i>	<i>Grade^a</i> (median)	<i>Crustal abundance^b</i>	<i>Clarke value</i>
Cu		Porphyry (volcano)	0.54%	27 ppm	200



^a All percentages are in weight percent. ^b Continental crust; see Chapter 3.01. ^c Highly variable. ^d In concentrate.



The Clarke value is the ratio of the concentration of an element in an ore, relative to its average crustal concentration.

Simon, 2015

Economic Geology: From Background to Ore

<i>Element</i>	<i>Oxidation state</i>	<i>Deposit type</i>	<i>Grade^a</i> (median)	<i>Crustal abundance^b</i>	<i>Clarke value</i>
Cu		Porphyry	0.54%	27 ppm	200
Na		Halite	40%	2.3%	17
Zn		Sedimentary exhalative	5.6%	72 ppm	780
As		Sulfide deposits ^c	~0.1%	2.5 ppm	~400
Rb		Lepidolite ^d	Up to 3%	49 ppm	~610
Mo		Climax	0.19%	0.8 ppm	2,400
W		Skarn	0.66% WO ₃	1 ppm	6,600
Pb		Sedimentary exhalative	2.8%	11 ppm	2,500
V		Layered mafic intrusions	~0.6%	138 ppm	~43
Au		Veins/Homestake	~10 ppm	1.3 ppb	~7,700
Ag		Creed vein	125 ppm	56 ppb	2,200
Ni		Komatiite	1.5%	59 ppm	250

^a All percentages are in weight percent. ^b Continental crust; see Chapter 3.01. ^c Highly variable. ^d In concentrate.

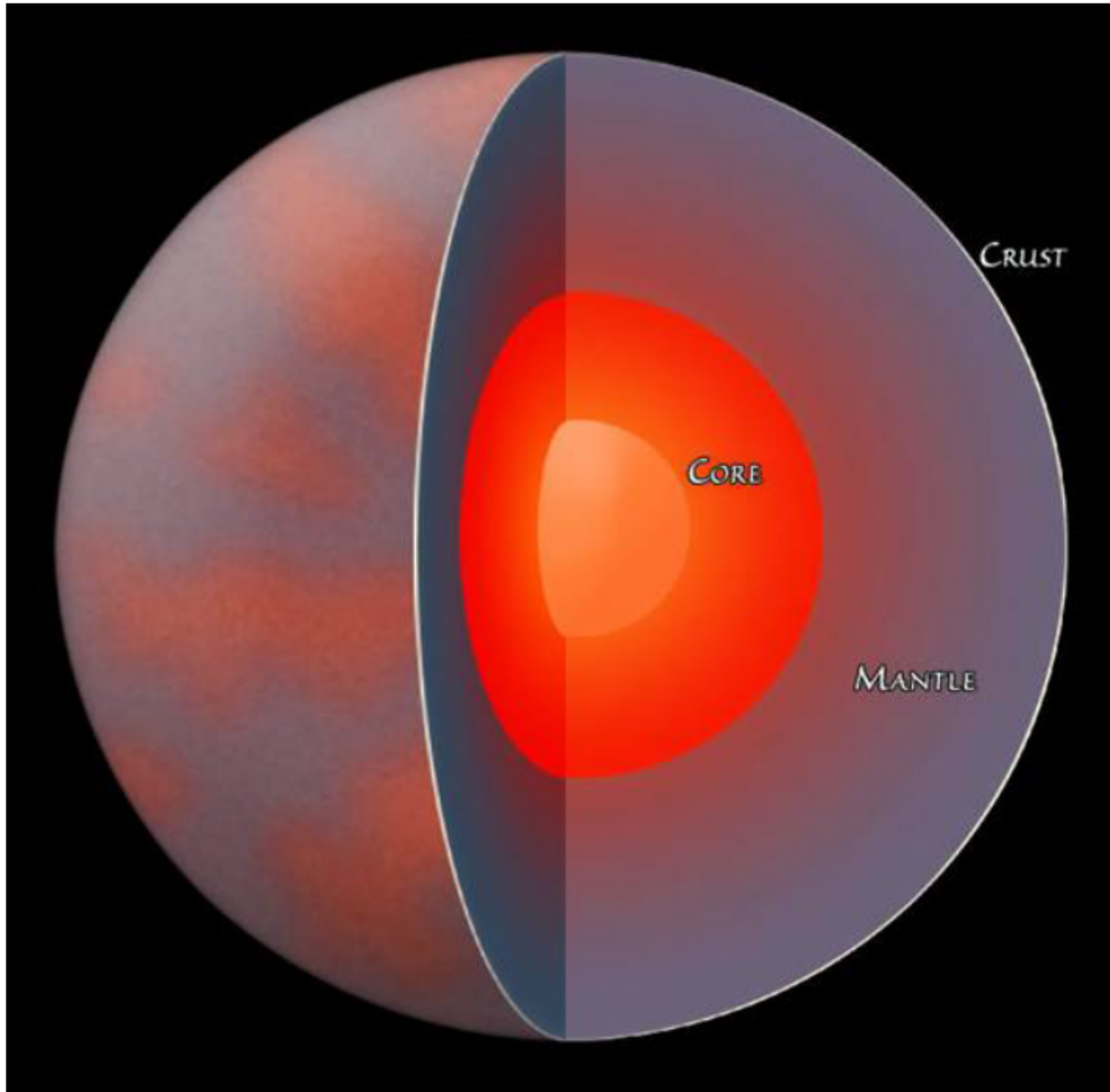


The Clarke value is the ratio of the concentration of an element in an ore, relative to its average crustal concentration.

Simon, 2015

ORIGIN OF MINERAL DEPOSITS - DIFFERENTIATION

Composition of Earth



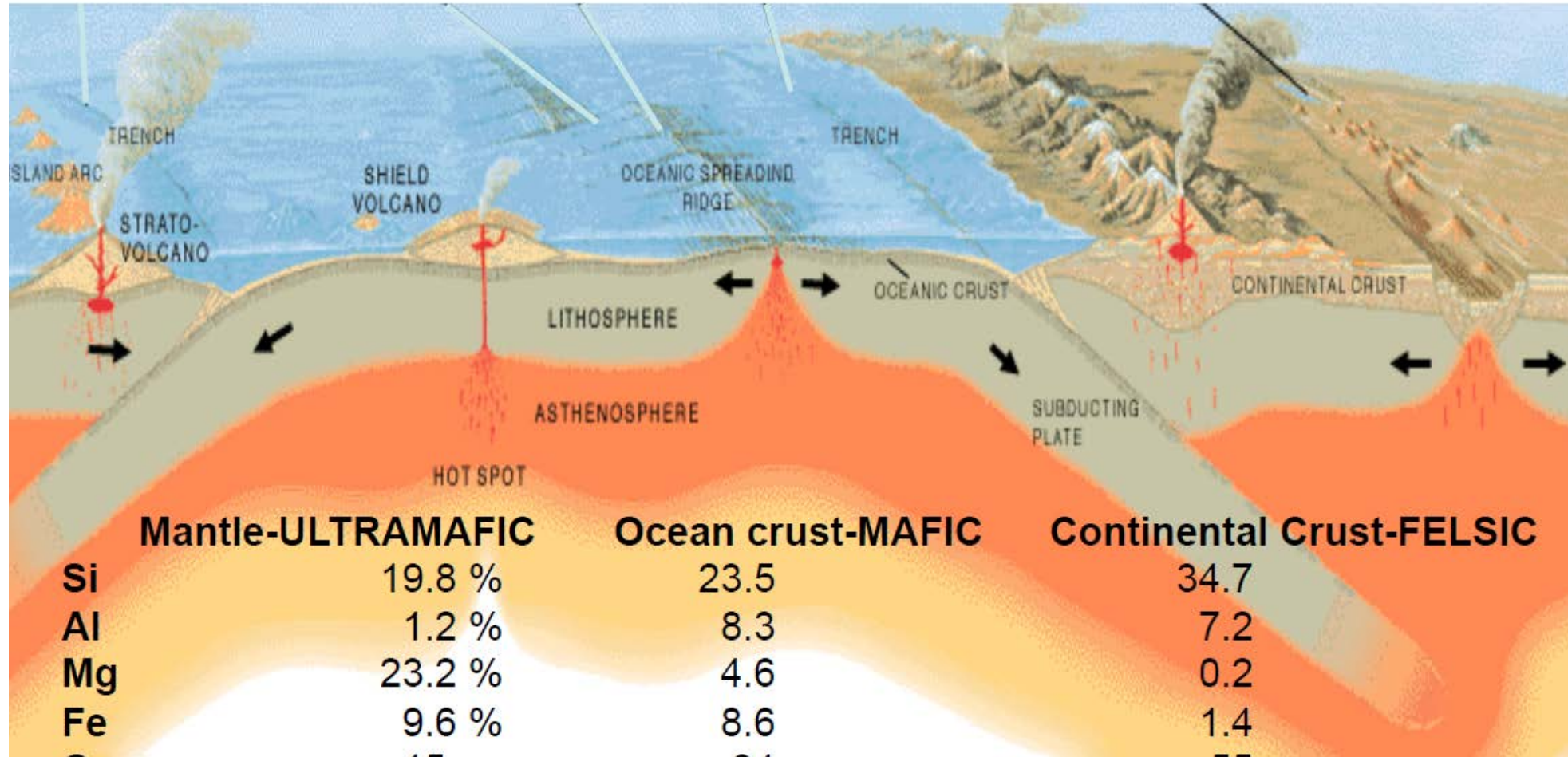
	<u>Crust</u>	<u>Earth</u>
<i>Major Elements</i>		
O	46.6	36.0
Si	27.7	15.0
Al	8.1	1.1
Fe	5.0	35.0
Ca	3.6	1.1
Na	2.8	0.6
K	2.6	0.1
Mg	2.1	13.0
Ti	0.4	0.1
H	0.1	-----
<i>Trace Elements</i>		
Mn	0.009	0.022
as ppm (parts per million)		
Sn	2	0.25
Cu	28	60
Ni	58	24000
Nb	20	1

Note how elements

*concentrate in
different reservoirs*

ORIGIN OF MINERAL DEPOSITS - DIFFERENTIATION

Composition of Mantle, Ocean Crust and Continental Crust



	Mantle-ULTRAMAFIC	Ocean crust-MAFIC	Continental Crust-FELSIC
Si	19.8 %	23.5	34.7
Al	1.2 %	8.3	7.2
Mg	23.2 %	4.6	0.2
Fe	9.6 %	8.6	1.4
Cu	15 ppm	94	55
Pb	0.5 ppm	7	15
Sn	0.5 ppm	1.5	3
Ni	2000 ppm	145	4.5

Simon, 2015

ORIGIN OF MINERAL DEPOSITS

Geologic processes that form mineral deposits (by forming ore minerals) along with deposits and elements that form by each process.

Surface Processes

Physical Sedimentation

Flowing water

Placer deposits - gold, platinum, diamond,
(stream or beach) ilmenite, rutile, zircon, sand, gravel

Wind

Dune deposits - sand

Chemical Sedimentation

Precipitation from
or in seawater

Evaporite deposits - halite, sylvite, borax, trona
Chemical deposits - iron, volcanogenic massive

Organic Sedimentation

Organic activity or
accumulation

Hydrocarbon deposits - oil, natural gas, coal
Other deposits - sulfur, phosphate

Subsurface Processes

Involving Water

Groundwater and related deposits - uranium, sulfur

Basinal brines - Mississippi Valley-type, sedex

Seawater - volcanogenic massive sulfide, sedex

Magmatic Water - porphyry copper-molybdenum, skarn

Metamorphic water - gold, copper

Involving Magmas

Crystal segregation - chromium, vanadium

Immiscible magma separation - nickel, copper,

cobalt, platinum-group elements

Exploration



Exploration 101

- Decide on a commodity/ metal(s) of interest
 - What deposit type?
 - How does it form?
- What data will be most useful for finding the deposit?
 - Geochemistry?
 - Geophysics?
- Are there data available that lead you to a specific geographic area?
 - Where (country, region) do you want to be?

Geochemistry

- What elements will be most useful for finding the deposit?
- What samples can you collect?
 - Soil (stream/ lake sediments)
 - Rocks
 - Minerals
 - Stream water
 - Plant material

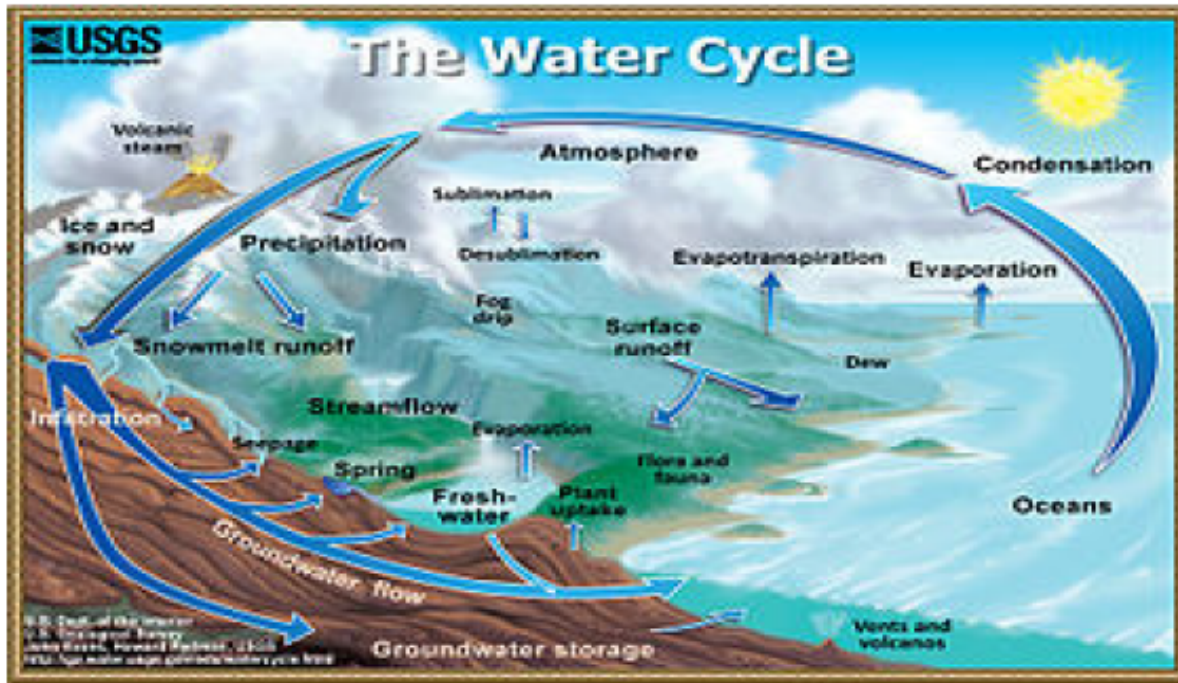
Geochem Targeting Tool Box

- Deposit models predict mineralogy, zonation and alteration
- Pathfinder elements or alteration zones can provide a larger target than the mineralization
- Element associations also depend on magma fractionation, weathering profile, sample type and analytical method

Style Of Deposit	Element Association*
Ni (Cu, PGM) Sulphides	Ni-Cu-PGM-(Cr-Se-Bi-Co-Au)
Porphyry CuAu, CuMo, Mo	Cu-Au-Mo-(Zn-Pb-Ag-Sb-Ag)
Epithermal AuAg	Au-Ag-As-(Cu-Pb-Zn-Sb-Hg-W)
Skarns	Cu-Au-Zn-Pb-W-Mo-Sn-Bi-As-Sb
Proterozoic Fe-CuAu	Cu-Au-U-Co-(REE-Ba-F-Sn-W-Mo-Bi-As)
VHMS	Cu-Pb-Zn-Ag-Au-(As-Sb-Sn-Bi-Mo-Ba-W)
SHMS	Cu-Pb-Zn-Ag-(As-Sb-Mo-Mn-Bi-Au)
MVT	Pb-Zn-(Cd-Ag-Ba-F-Cu-Ni)
Red Bed Cu	Cu-Ag-Co-(Pb-Zn-U-Se-Mo)
Sedimentary UV(Cu)	U-V-Au-Mo-Se-(Ni-As-Co-Cu-Pb-Zn)
Mesothermal Au	Au-As-(Sb-W-Cu-Pb-Zn-Hg)

*In the weathering environment, some of these elements will be present in resistate minerals e.g. cassiterite and chromite.





Primary Geochemistry

Element distribution in rocks associated with deposit formation

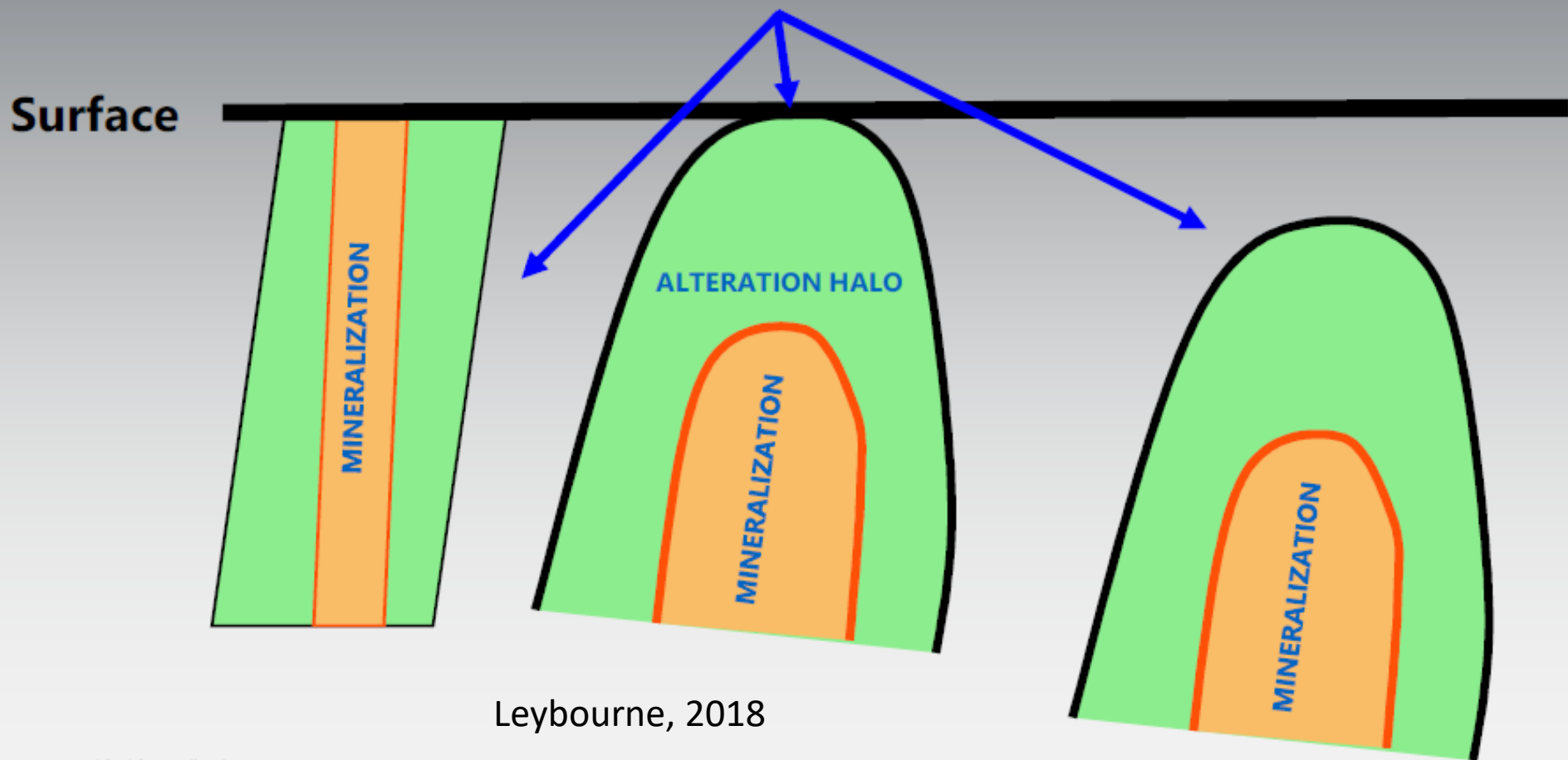
Secondary Geochemistry

Element distribution in the geochemical cycle (water, vegetation, soils, streams, glacial sediments)

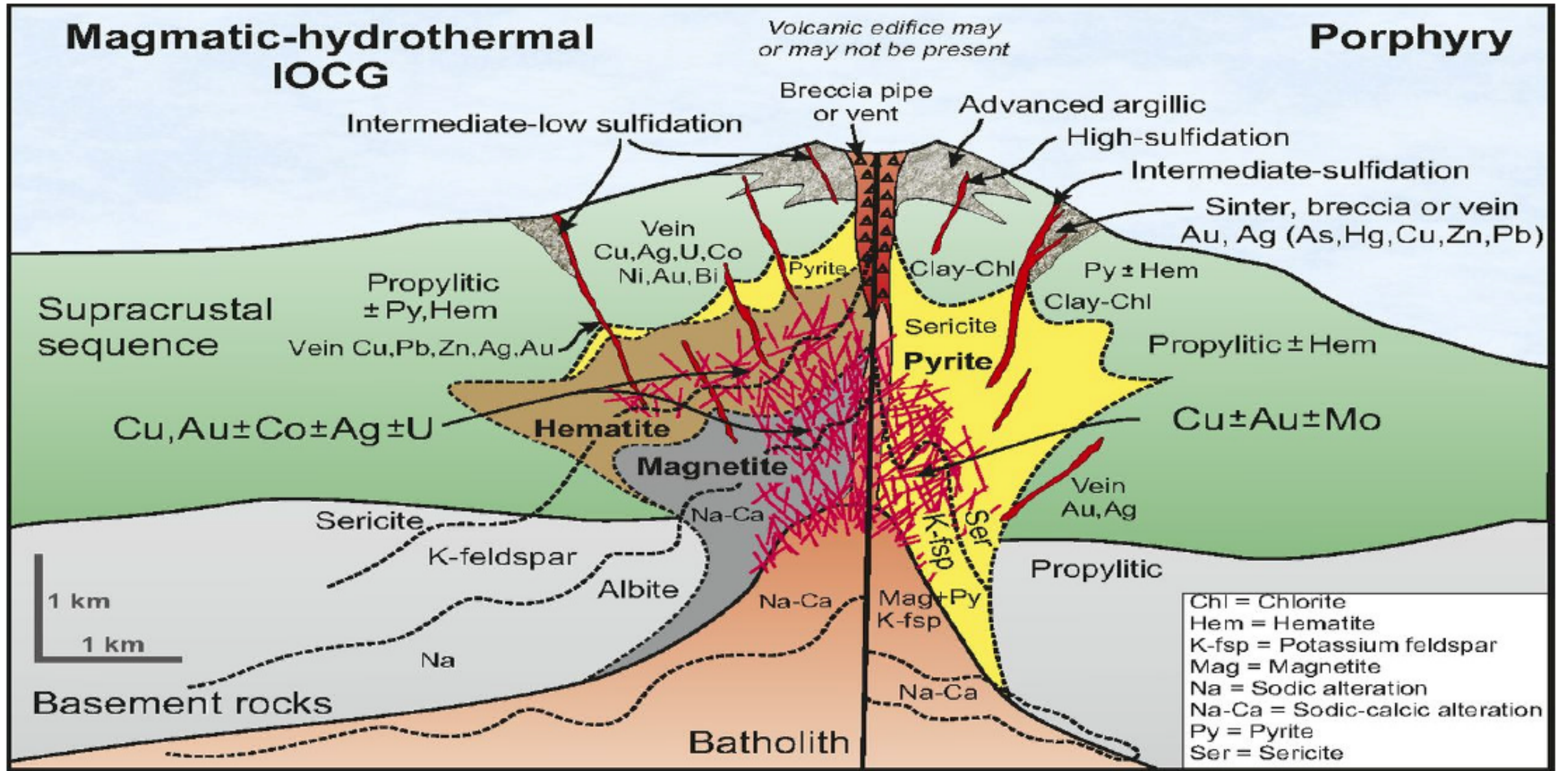
Dispersion mechanically or by chemical mobility

Background

Zones of geochemical enrichment in target and/or pathfinder elements



Leybourne, 2018



Analytical methods

Choice of analytes to focus on for data analysis and interpretation – depends on the style of mineralization and the relative mobility by mechanical or hydromorphic dispersion

Type of deposit	Major components	Minor components	Labile components ^A	Relatively immobile components ^B
VMS	Fe, S, Cu, Zn, Pb	Cd, Hg, Au, As, Sb, Ba, Bi, In	Fe, S, Zn, Cu, As, Cd, Hg, Sb	Pb, Bi, In, Au, Ag, Ba
Porphyry Cu ± Mo	Cu, Mo, S	Fe, Ag, Au, Se, Re, As	Cu, Mo, S, Fe, Se, As, Re	Ag, Au
SEDEX	Fe, S, Cu, Zn, Pb	Ag, Au, Ba, Cd	Fe, S, Zn, Cu, Cd	Pb, Ba, Au, Ag
Gold (vein)	Au, Ag	As, Sb, Se, Te, S, Hg	S, Se, As, Hg, Te, Sb	Au, Ag
Ni-Ci-PGE	Ni, Cu, PGE	Cr, Co, S	Cu, S, PGE	Co, Ni, Cr
Kimberlite (diamond)	Sr, Nb, Ba, Cr, Ni	LILE, HFSE, REE	Sr, LILE	Ba, HFSE, Nb, Ba, Cr, Ni, REE
Unconformity uranium	U	Se, Mo, V, Cu, Pb	U, Se, Cu, Mo	U, Pb, V

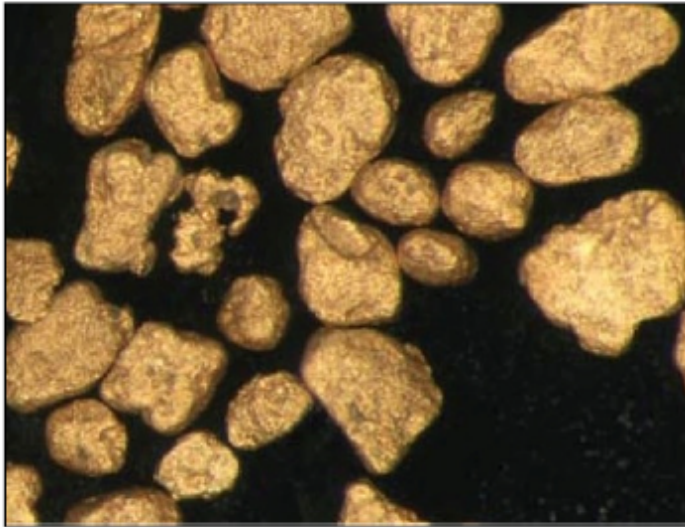
Gold!!!!

Gold deposits

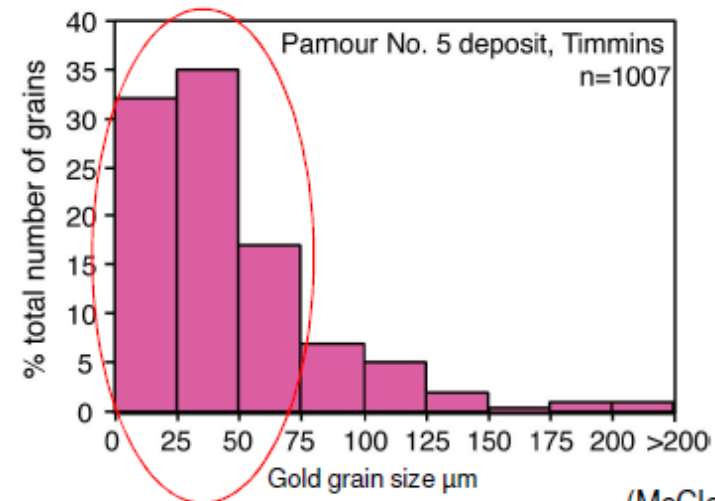
- Greenstone hosted/ orogenic deposits
- Reduced intrusion related gold deposits
- Porphyry, epithermal and skarn deposits
- Iron oxide copper gold deposits
- Gold rich volcanogenic massive sulphide deposits
- Placer gold deposits



Gold Indicator Minerals



- Recovered from stream sediments since Roman times
- Most well known and widely used indicator mineral
- Other indicator minerals include sulphides, arsenides, tellurides, scheelite, tourmaline, rutile, barite, secondary minerals (jarosite, limonite, goethite, pyrolusite)
- Gold grains easily recovered from 10 μm to 2 mm size fraction of sediments
- Most gold grains recovered from till, 10 to 75 μm
- Gold within sulphides or in solid solution with sulphides will be detected with automated mineralogy methods and matrix till geochemistry



McClenaghan, 2018

(McClenaghan, 1999, 2001)

Gold Grain Morphology

- DiLabio (1990) classification scheme describes conditions and surface textures of gold grains related to glacial transport distance

- **Pristine gold grains:**

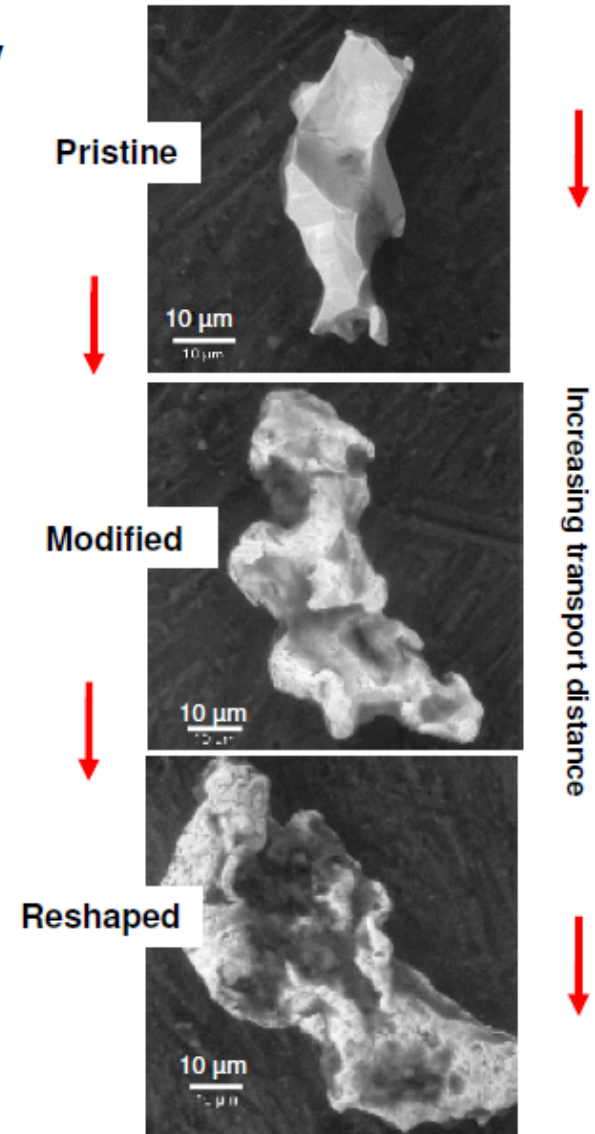
- Primary shapes and surface textures
- Appear not to have been damaged in transport
- Angular wires, rods and delicate leaves that once filled in fractures, occurred as crystals with grain molds, and as inclusions in sulphides

- **Modified gold grains:**

- Some primary surface textures
- Edges and protrusions have been damaged during transport
- Commonly striated
- Irregular edges and protrusions are crumpled, folded and curled
- Grain molds and primary surface textures preserved on protected faces of grains

- **Reshaped gold grains:**

- Primary surface textures destroyed
- Original grain shape no longer discernible
- Flattened to rounded resulting from folding of leaves, wires, rods
- Surfaces may be pitted from impact marks from other grains
- Surfaces are not leached of silver in most cases in glaciated terrain



(McClenaghan, 2001)

McClenaghan, 2018

Gold geochemistry

Economic Geology
Vol. 94, 1999, pp. 649-664

Lode and Placer Gold Composition in the Klondike District, Yukon Territory, Canada: Implications for the Nature and Genesis of Klondike Placer and Lode Gold Deposits

J. B. KNIGHT,[†]

Micro Science, RR #1, Site 29, Comp 10, Smithers, British Columbia, Canada V0J 2N0

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*Department of Ocean and Earth Sciences, 6339 Stores Rd., University of British Columbia, Vancouver,
British Columbia, Canada V6T 2B4*

AND S. R. MORISON

Gartner Lee Ltd., Unit 212, 212 Main St., Whitehorse, Yukon Territory, Canada Y1A 2A9

Microchemical Characterization of Alluvial Gold Grains as an Exploration Tool

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Mike Styles***

School of Earth Sciences, University of Leeds, Leeds
LS2 9JT, United Kingdom

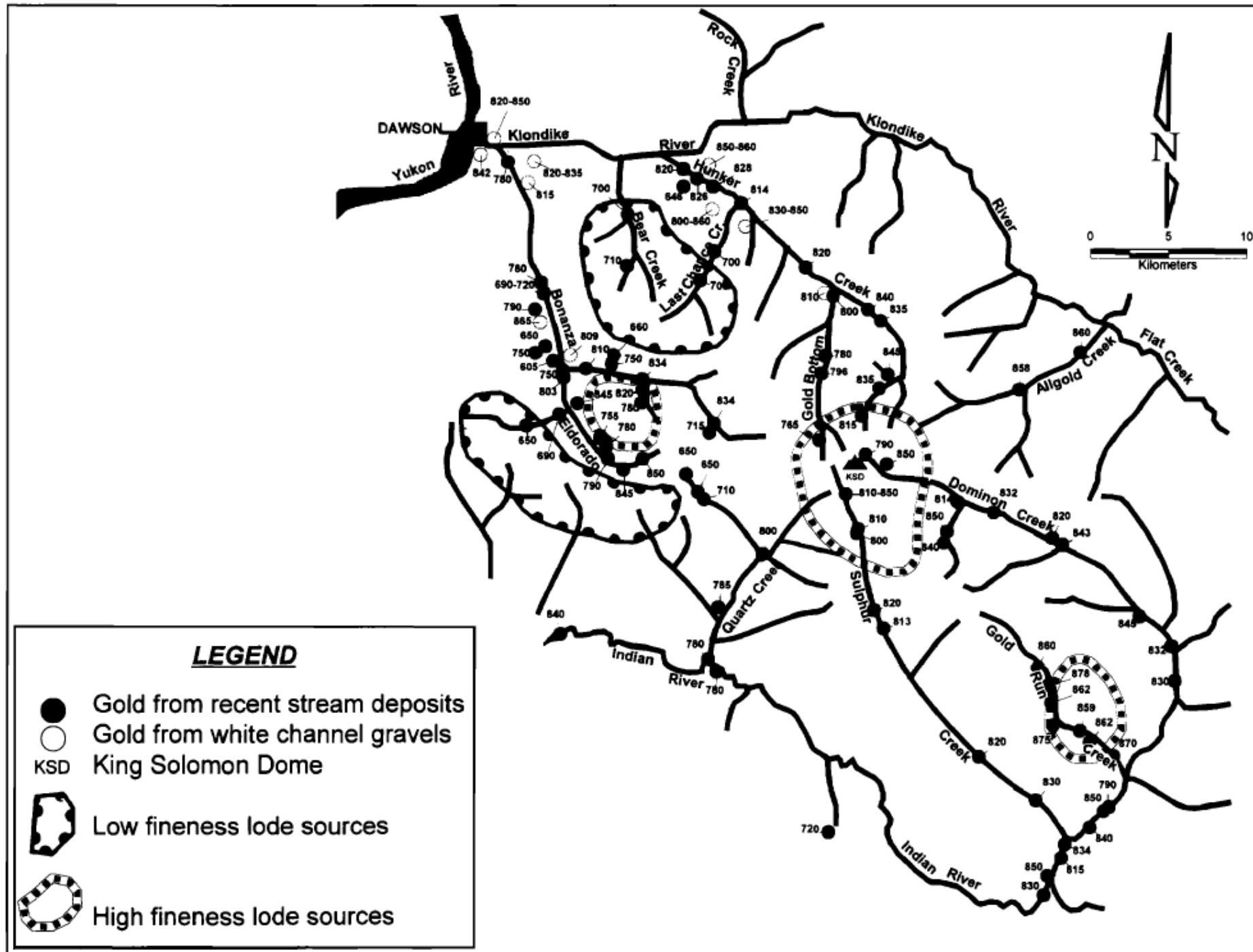
*British Geological Survey, Keyworth, Notts NG12 5GG,
United Kingdom

E-mail: aedrjc@LUCS-03.NOVELL.LEEDS.AC.UK

Received: 15 October 2001

Application of microchemical characterization of placer gold grains
to exploration for epithermal gold mineralization in regions
of poor exposure

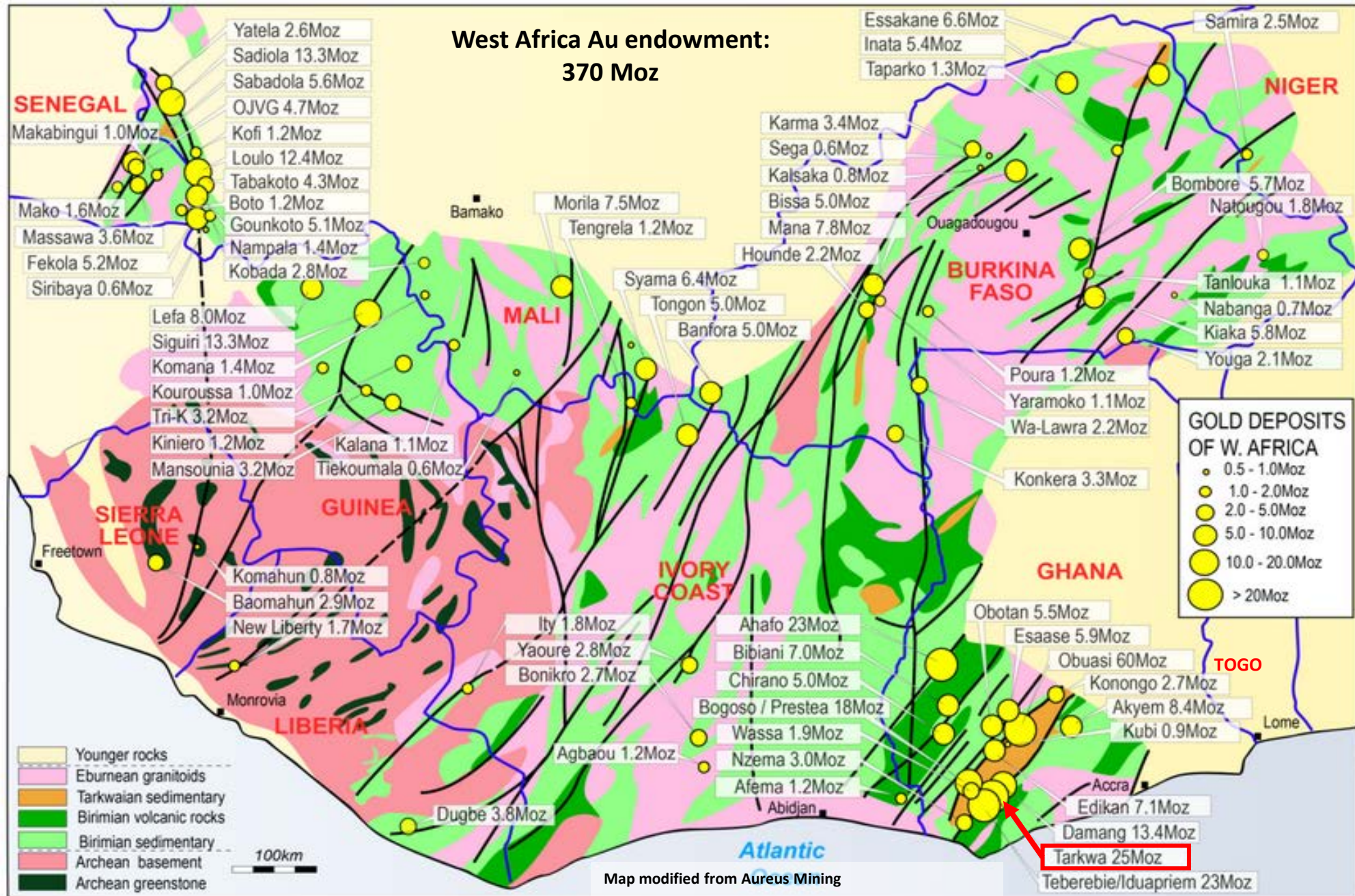
R.J. Chapman ^{a,*}, J.K. Mortensen ^b



Knight et al., 1999

Gold in Ghana....

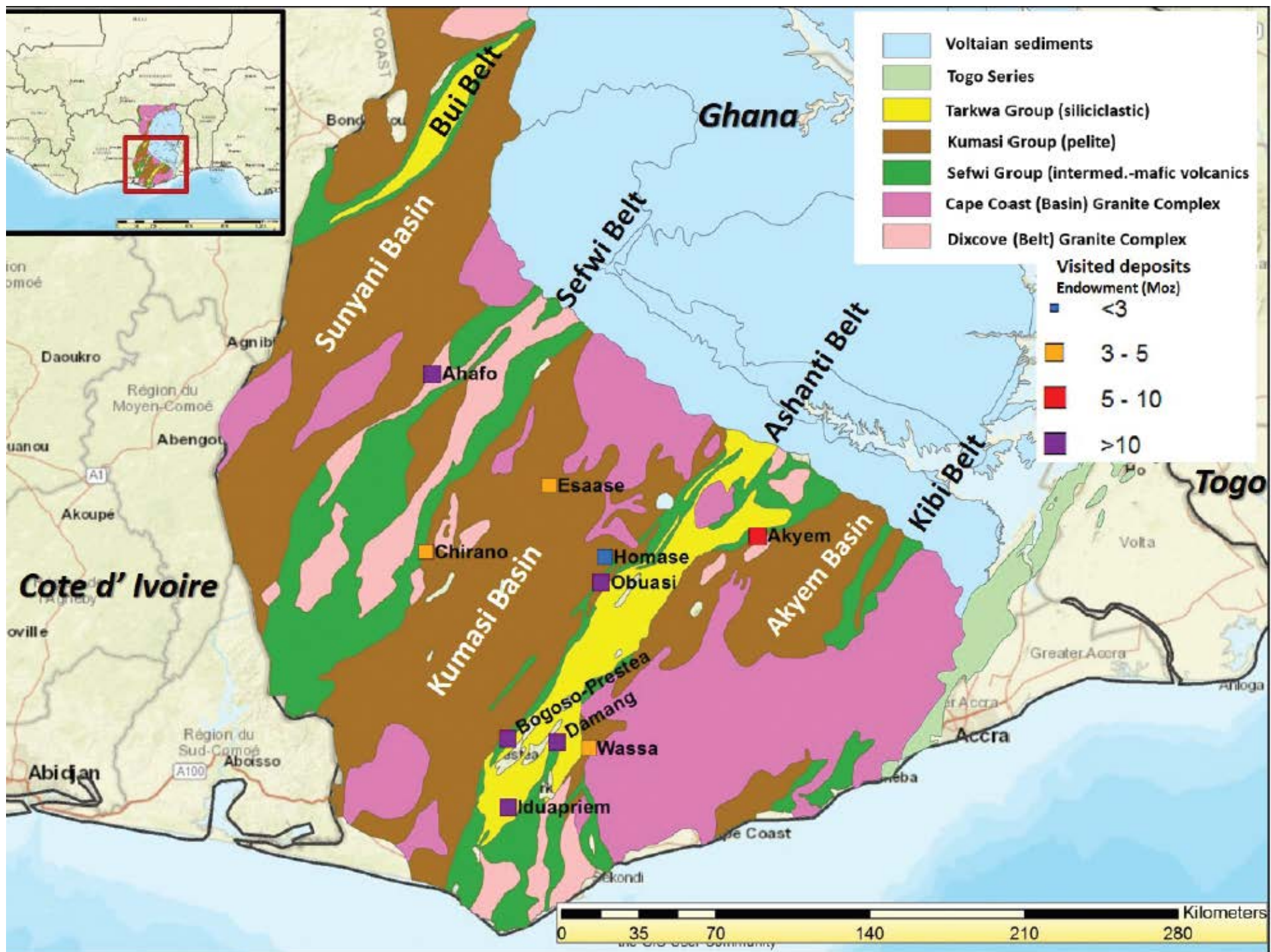
Regional geology and gold deposits



Gold Ore from the Tarkwa basin



Visible gold associated with pyrite from orogenic gold deposits of the Birimian shield in Ghana



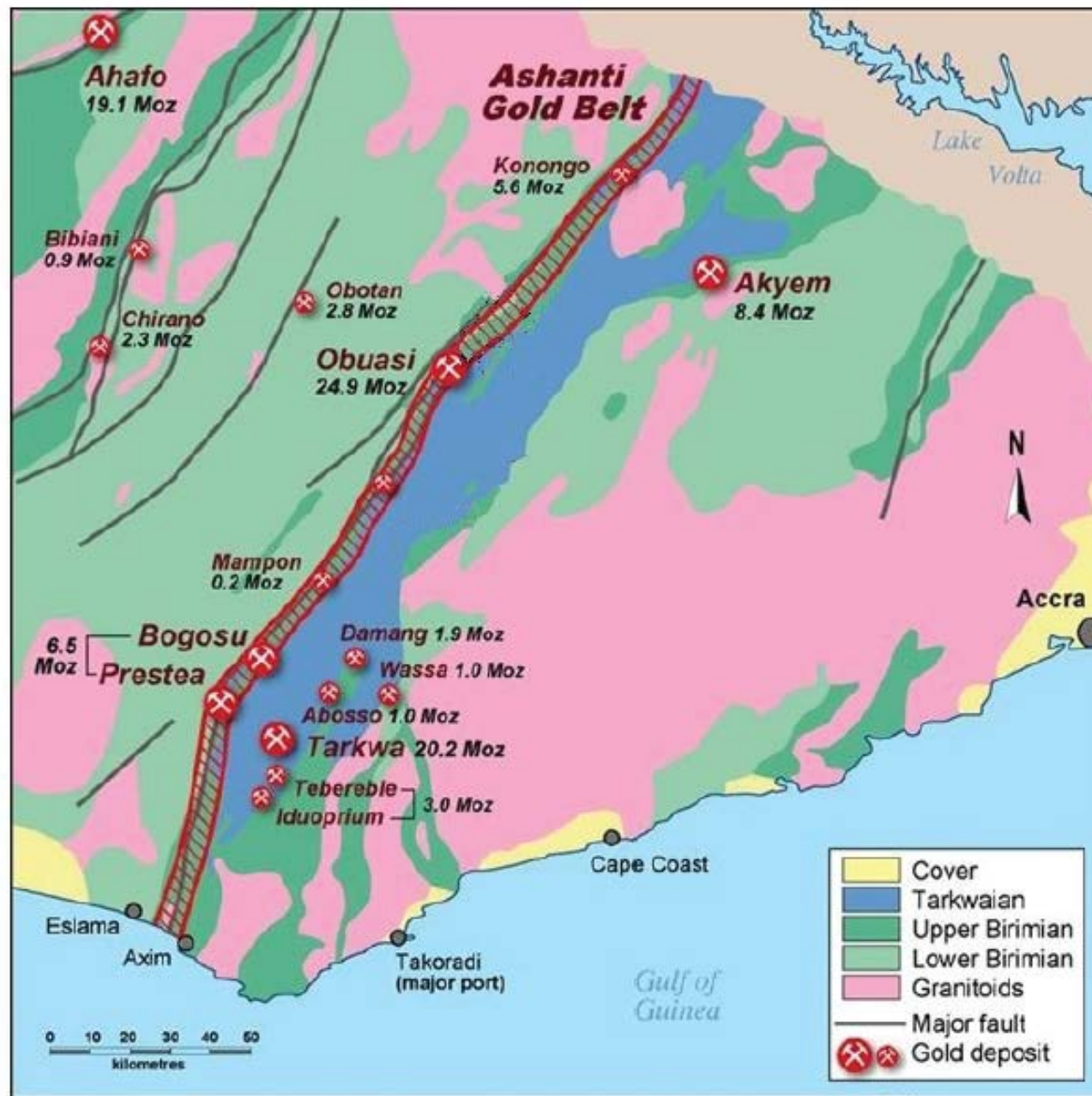


Figure 6-7. Location of Obuasi gold mine along the Axim-Konongo shear zone along the northwestern margin of the Ashanti greenstone belt. Other significant mines are also shown (Metals News, 2008).



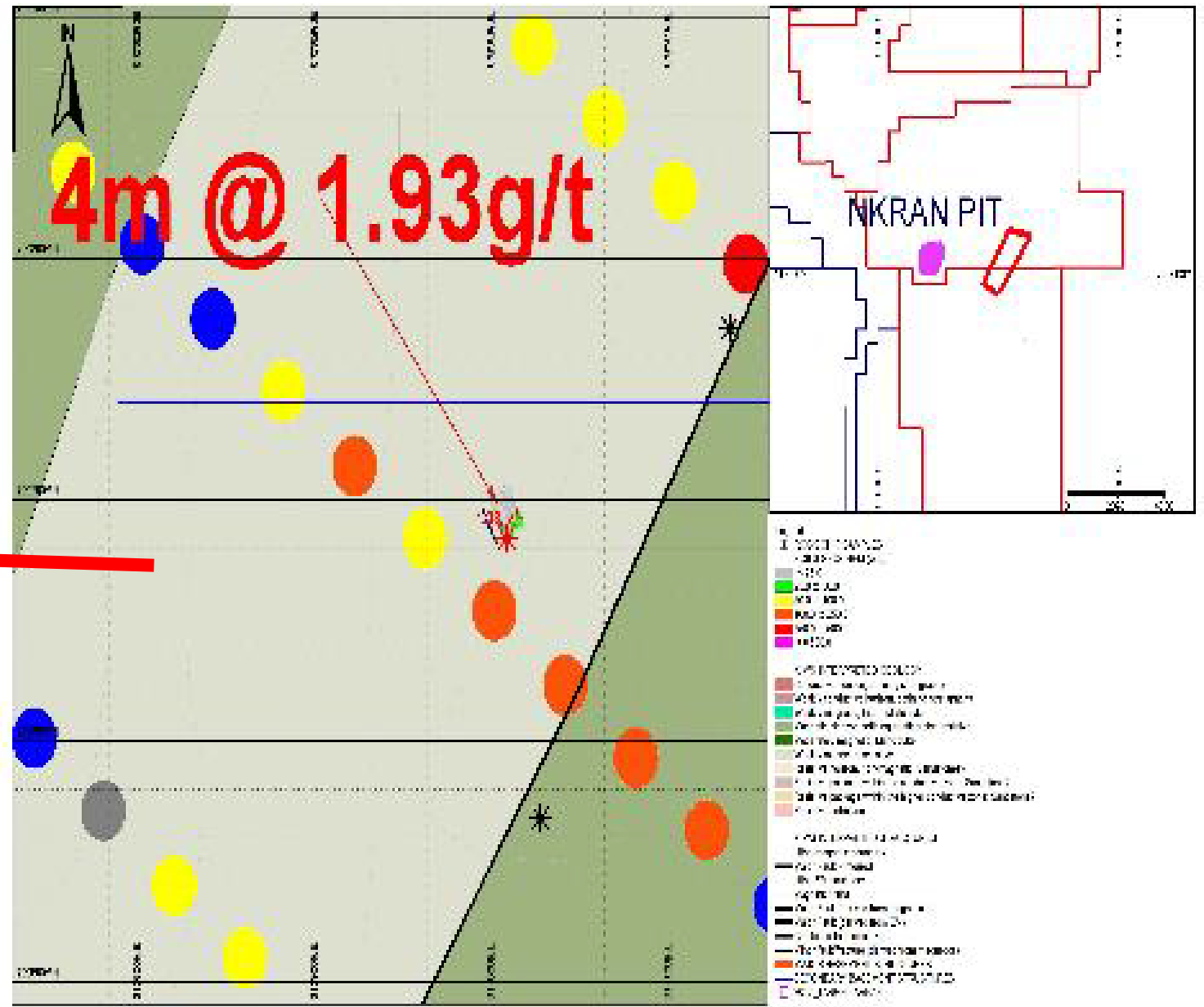
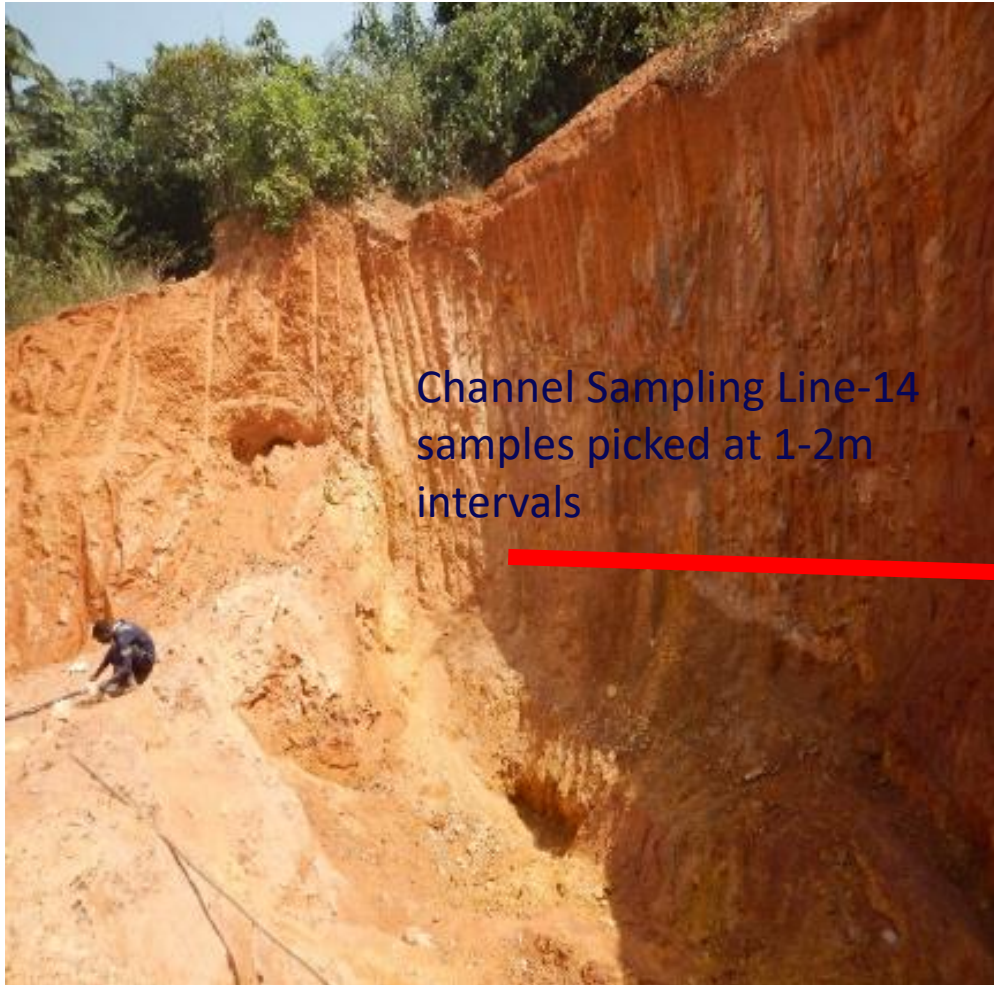
Ashanti belt main structure



Mined out pit on the Ashanti main structure

Using geochem for exploration in Ghana

CHANNEL SAMPLING



AsankoGold, 2018

Structure and Mineralization Characteristics

▪ **Structure**

- Conforms with the general regional strike NNE – SSW structure, with folds plunging to NE
- Multiplicity of deformational episode

▪ **Mineralisation**

- Complex arrangement of deformed rocks
- Mineralisation is hosted in deformation zones and proximal to contacts
 - ✓ Styles of mineralization
 - ❖ Disseminated Arsenopyrite domains (Refractory)
 - ❖ Contact-related Vein Quartz
 - ❖ Granitoid/Stockworks (primary non- refractory)
 - ❖ Supergene oxide and transition ore
 - ✓ Types of mineralization
- Quartz vein type (quartz with free Gold, in association with lesser amount of various metal sulphides containing iron, zinc, lead and copper), non refractory
- Sulphide ore (inclusion of gold in the crystal structure of arsenopyrite minerals. Higher gold grades tend to be associated with finer grain arsenopyrite crystals. Sulphide ore is generally refractory)

Two types of Mineralization



Sulphide Mineralisation with ASP



Quartz with Spec of Gold

Vary from 1-40m wide and 1-1,000g/t Au.

Associated Attributes:

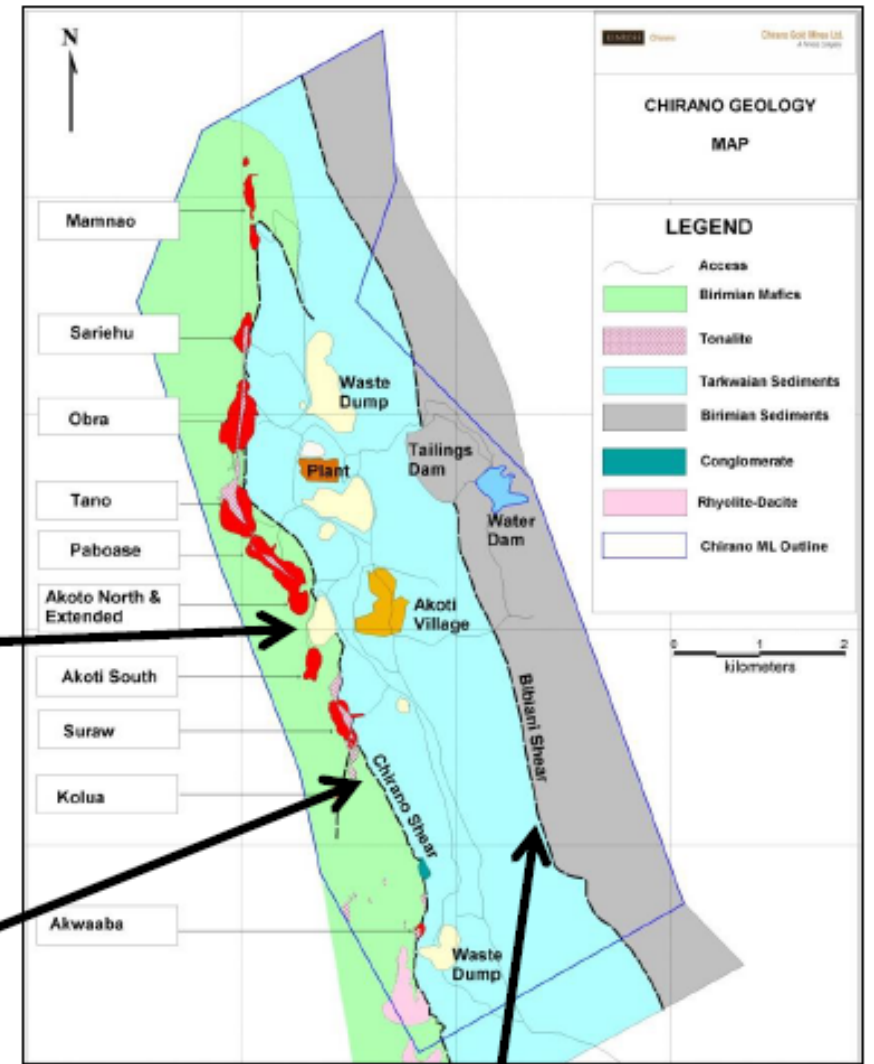
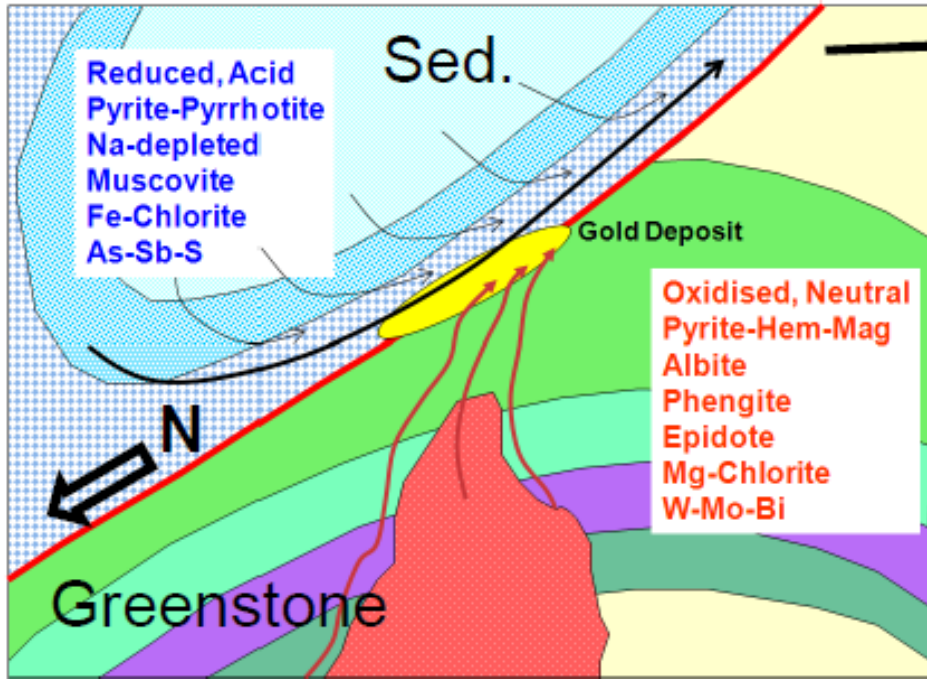
Strong shearing, potentially faulted, fractured and friable domains

Deposition mechanism:

Chirano gold system comprises two types of shear hosted :

1. Chirano: *Porphyry intrusion-mafic volcanic-hosted deposit*
2. Bibiani: *Sediment-hosted deposit*

- Different mineralogy and chemistry
- Fluid mixing (evidence pyrite deposition of graphite selvages in black breccia - Akwaaba)



Chirano:

- Hematite-magnetite-pyrite
- Intense albite-ferroan dolomite alteration
- W-Mo-Bi pathfinder signature.
- Phengitic sericite and Mg-rich chlorite
- Arsenic depletion
- Gold is strongly related to pyrite

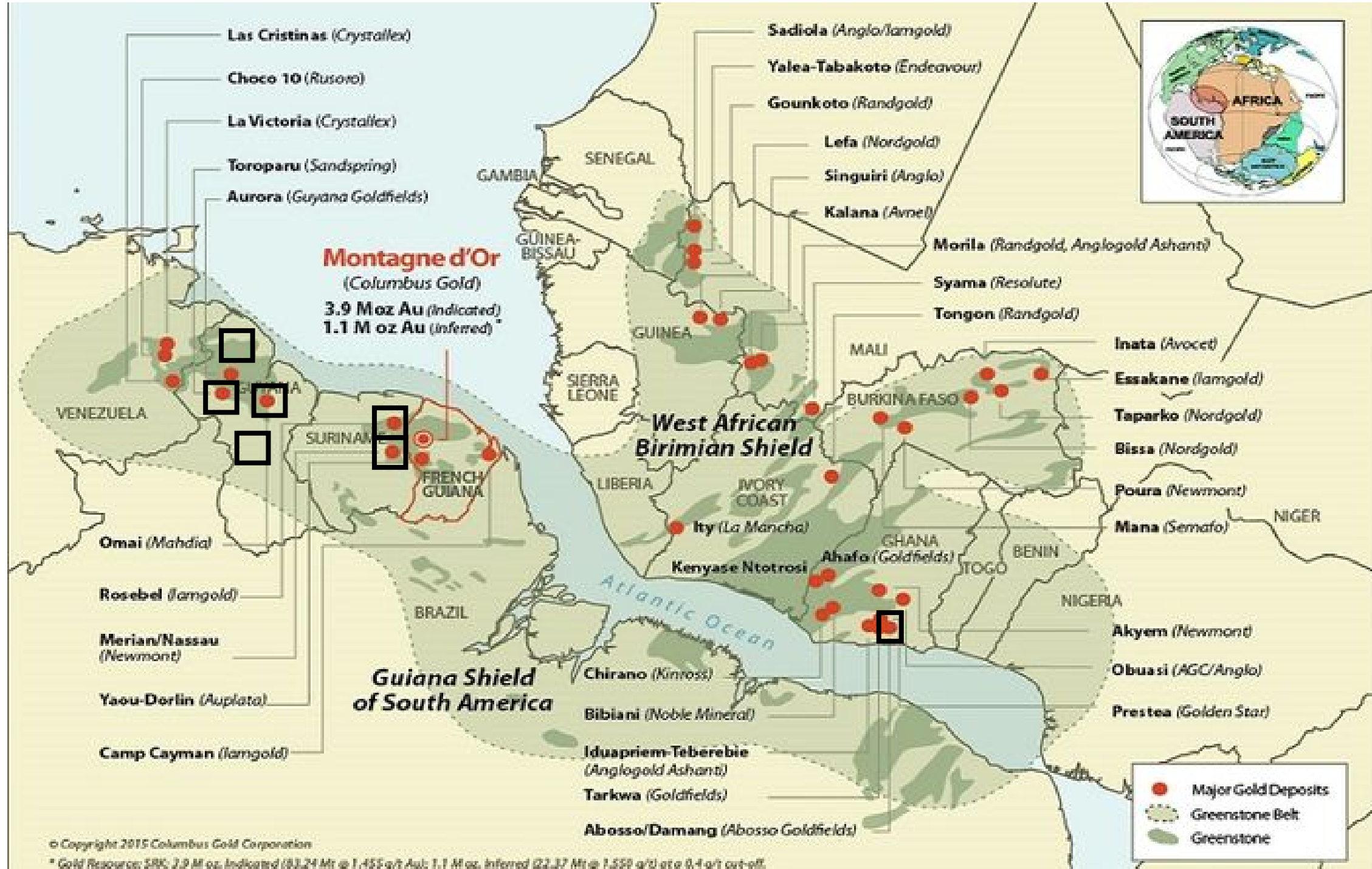
Bibiani

- Style graphitic shear
- Pyrite-arsenopyrite-graphite
- Intense sericite (paragonite)
- As-Sb pathfinder signature



- Prestea Underground is hosted within Birimian Phyllites
- Gold mineralization is associated with the prolific fault zone referred to as the Ashanti Trend
- West Reef mineralization is hosted in a fault structure parallel to the Main Reef located ~250m in the hanging wall
- Gold occurs as free gold along carbonaceous partings within the quartz veins or with pyrite and arsenopyrite

Au geochem for exploration?



Common Indicator Mineral Suites

- Gold grains (Au)
 - Native copper (Cu)
 - Kimberlite indicator minerals
 - Platinum Group minerals (PGM)
 - Sulphide minerals
 - Metamorphosed VMS (e.g. gahnite, staurolite)
 - Magmatic Ni-Cu-PGE minerals
 - Porphyry Cu-Mo-Au minerals
 - Scheelite, wolframite (W)
 - Cassiterite (Sn)
 - Cinnabar (Hg)
 - Fluorite, topaz (F)
 - Uranium minerals
 - Rare earth element (REE) minerals
-
- May be recovered from same heavy mineral concentrate
 - Archived concentrates can be re-examined



Trace Element Compositions Minerals

Sulphides	Present in % Amounts	Present in Trace Amounts
Pyrite	Cu, Co, Ni, Zn, As , Pb	Pb, V, Sb, Se, Mn, Ag, Au, Bi, Mo Cd
Pyrrhotite	Ni	Co, Cu, Mn, V, Zn, Sn
Arsenopyrite	Co	Mn, Ni
Chalcopyrite	Mn, Zn	Ni, Co, Sb, Ag, Se, As, Sn, V, Mo
Sphalerite	Fe, Mn, Cu, Cd, Sb , Sn, Ag, As	Co, Ni, Bi, In, Ga, Tl
Galena	Bi, Ag, Sb , Se, As	Cu, Mn, Sn, Zn, Cd
Silicates	Present in Major Amounts	Present in Minor Amounts
Plagioclase	K, Sr, Ba, Rb, Ti, Mn	P, Ga, V, Zn, Ni
K-feldspar	Na, Ca, Ba, Sr, Rb, Ti	P, Pb, Li, Ga, Mn, Al, Ti, Fe, Mg, Ca
Amphibole	Ti, F, K, Mn, Cl, Rb, Zn, Cr, V, Sr, Ni	Ba, P, Cu, Co, Ga, Pb
Pyroxene	Al, Ti, Na, Mn, K, Cr, V, Ni, Cl, Sr	P, Cu, Co, Zn, Li, Rb
Muscovite	Al, K, Fe, Ti, Mg, Ca, Na, Rb, Li	Mn
Biotite	Ti, F, Ca, Na, Ba, Mn, Rb, Cl, Zn, V, Cr, Li, Ni	Cu, Sr, Co, P, Pb, Ga
Olivine	Ni, Mn, Ca, Al, Cr, Ti, P, Co	Zn, V, Cu, Sc

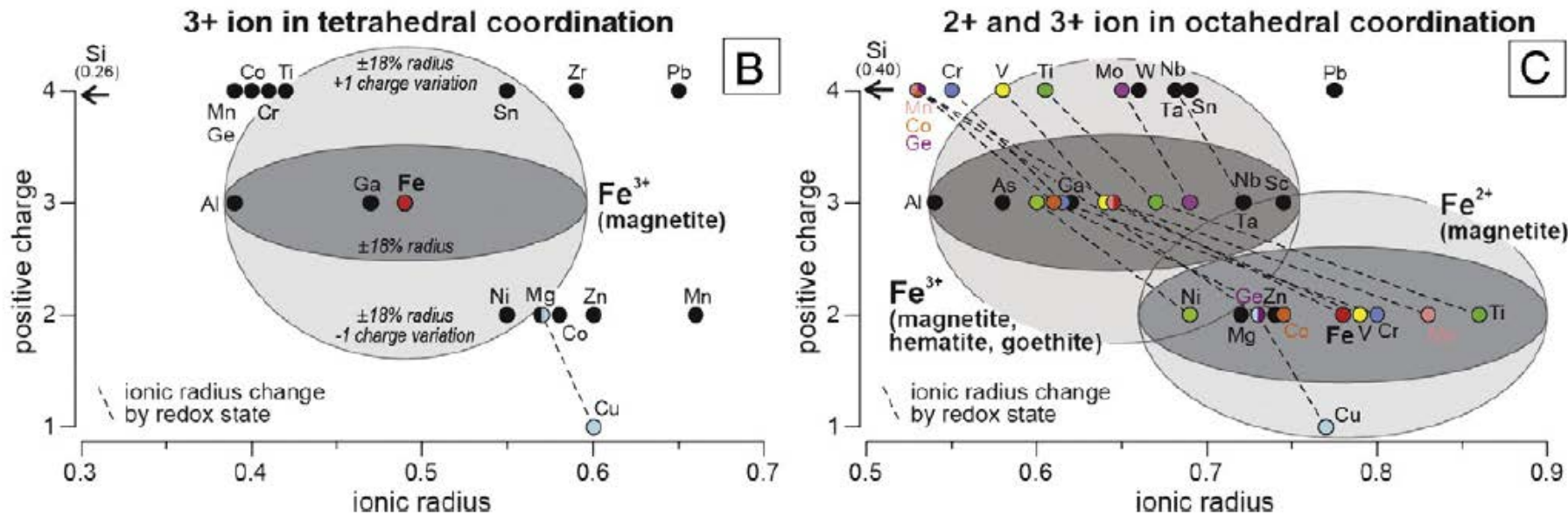
Chalcopyrite

*Data from Lelong et al (1976), Joyce (1984) and Vaughan and Craig (1978).

Mineral geochem as a tool for exploration

Why Magnetite?

- Iron oxide (Fe_3O_4)
 - Can incorporate other elements into its structure.



Why magnetite geochemistry?

DETRITAL MAGNETITE AS A PROVENANCE INDICATOR¹

JEFFRY D. GRIGSBY²

*Department of Geology
University of Cincinnati
Cincinnati, Ohio 45221*

Miner Deposita (2011) 46:319–335
DOI 10.1007/s00126-011-0334-y

ARTICLE

Discriminant diagrams for iron oxide trace element fingerprinting of mineral deposit types

Céline Dupuis • Georges Beaudoin

Ore Geology Reviews 61 (2014) 1–32



Contents lists available at [ScienceDirect](#)

Ore Geology Reviews

journal homepage: www.elsevier.com/locate/oregeorev



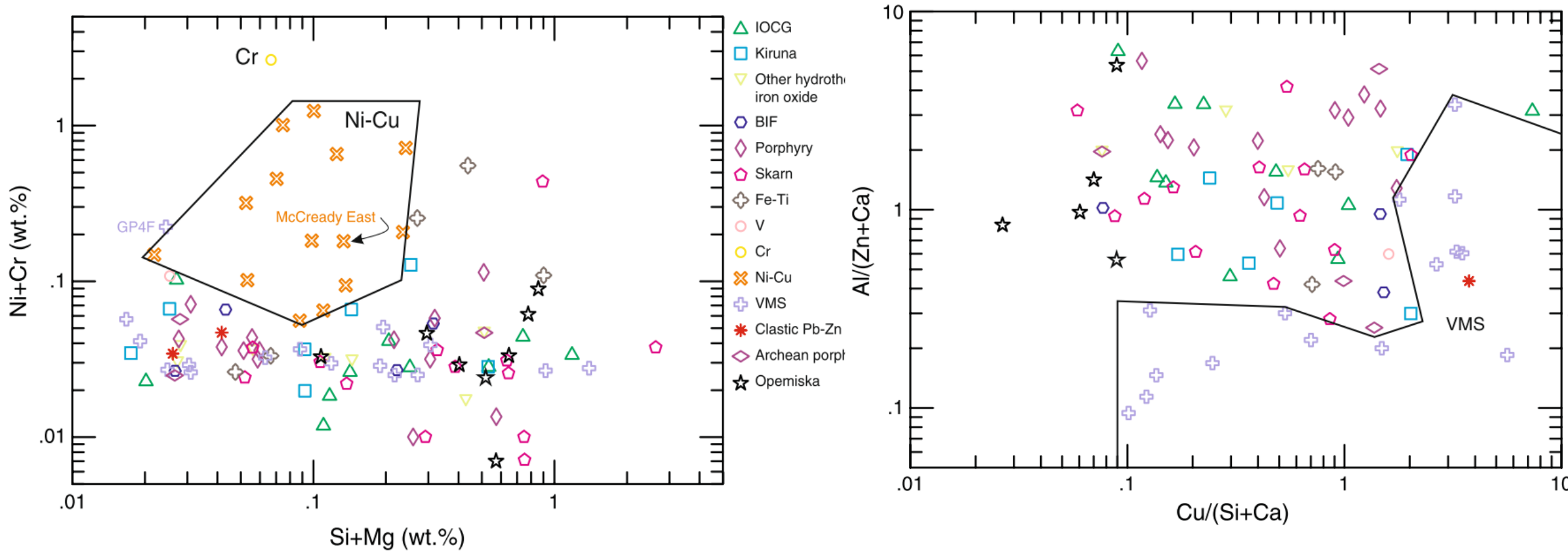
Review

The chemistry of hydrothermal magnetite: A review

Patrick Nadoll ^{a,*}, Thomas Angerer ^b, Jeffrey L. Mauk ^c, David French ^d, John Walshe ^a

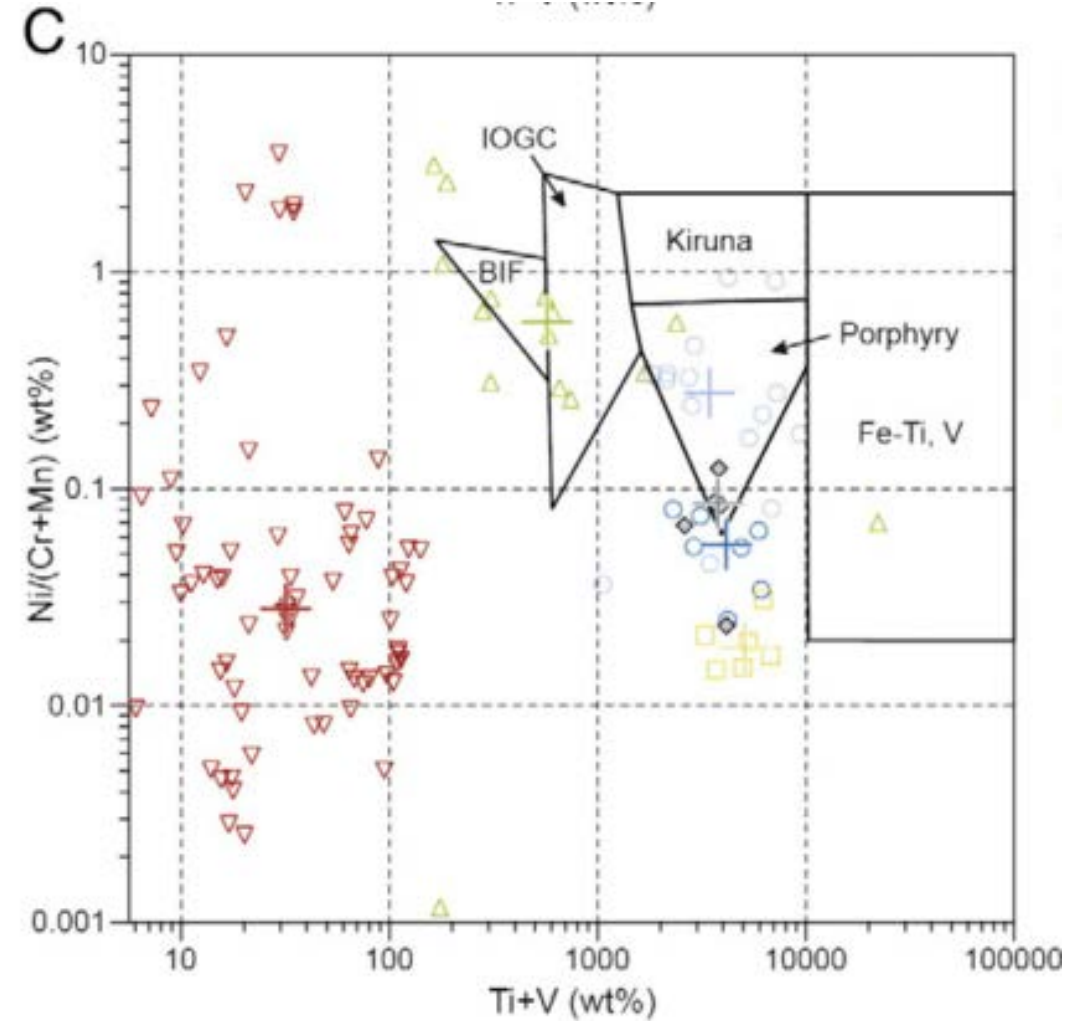
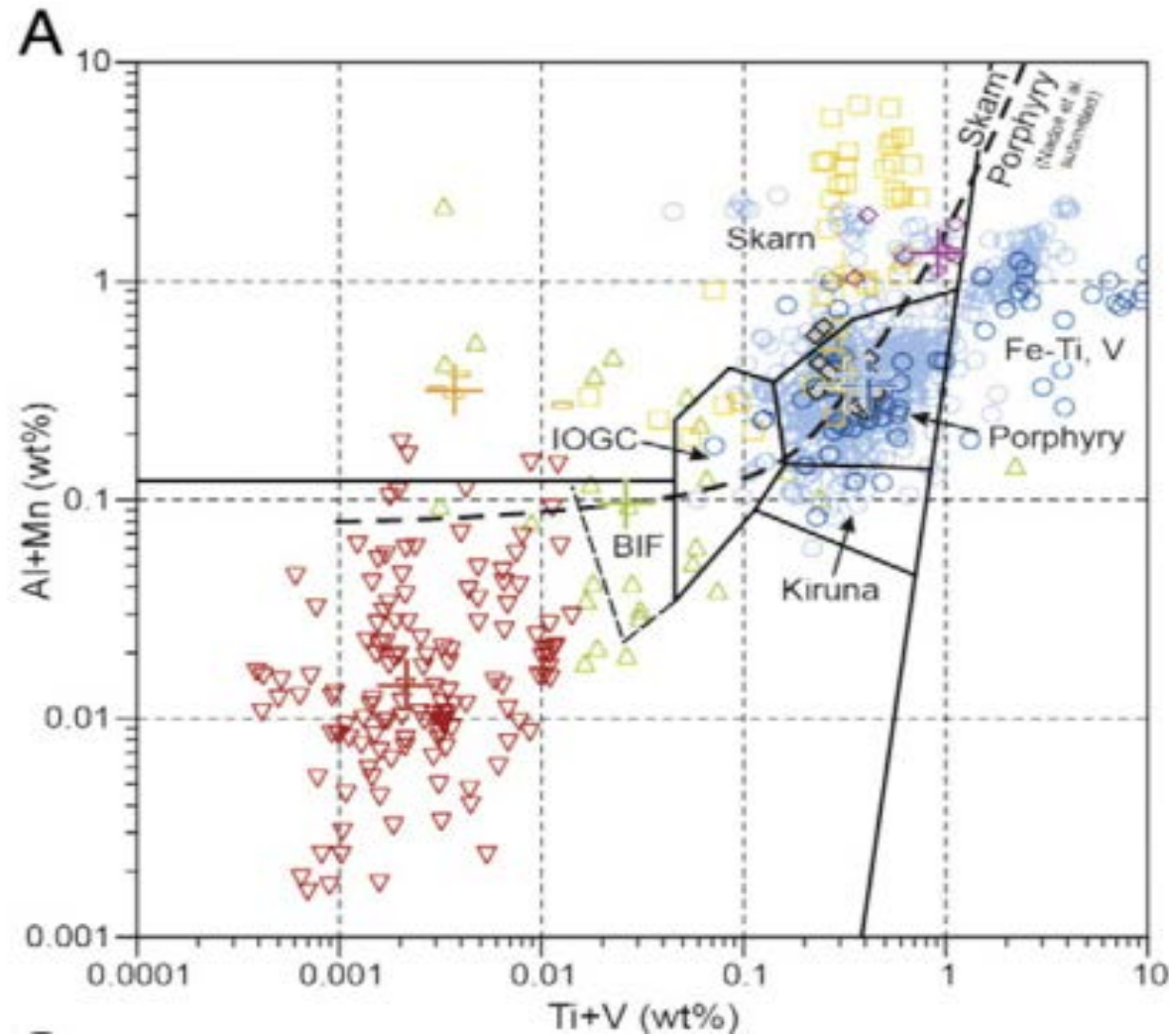


Variations in magnetite chemistry



Dupuis and Beaudoin, 2011

Variations in magnetite chemistry



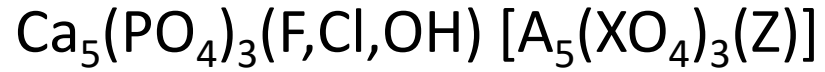
Guyana Magnetite project

- Collect magnetite grains
- Analyze magnetite grains
- Use magnetite discriminant plots to determine ore deposit potential



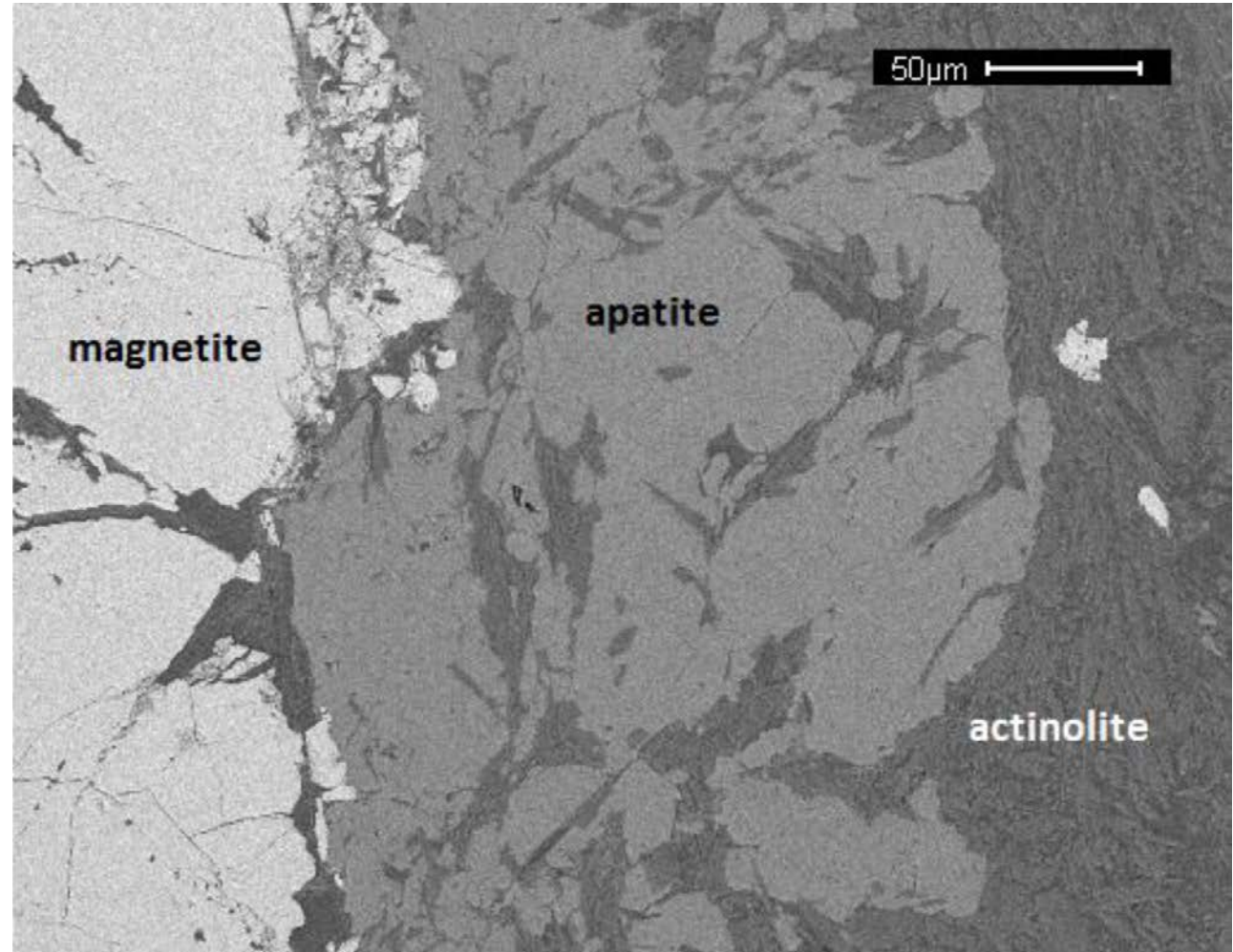
Mineral geochemistry for understanding how
ore deposits form

Los Colorados apatite chemistry?

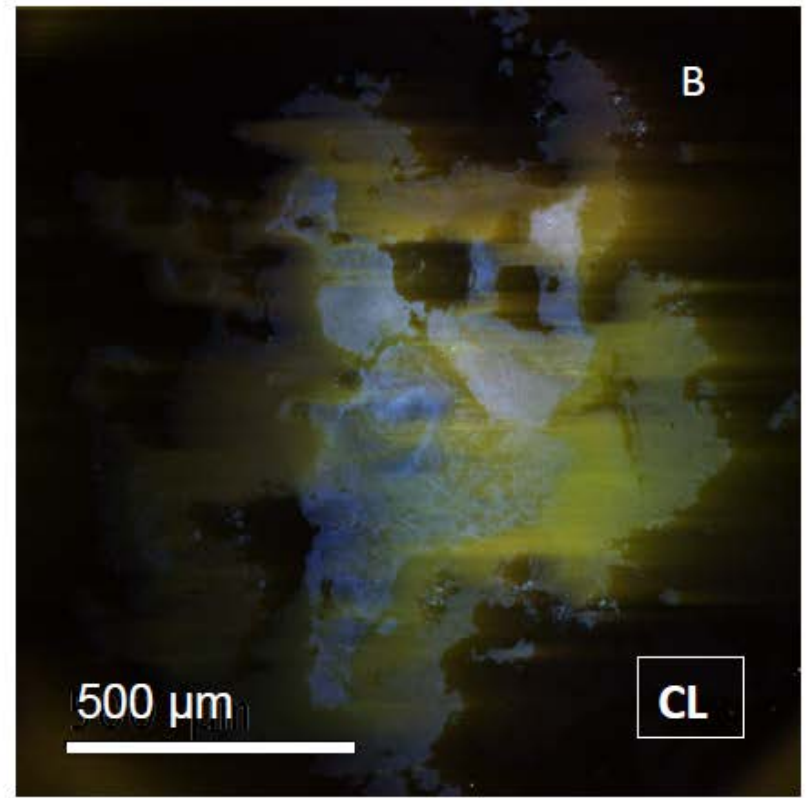
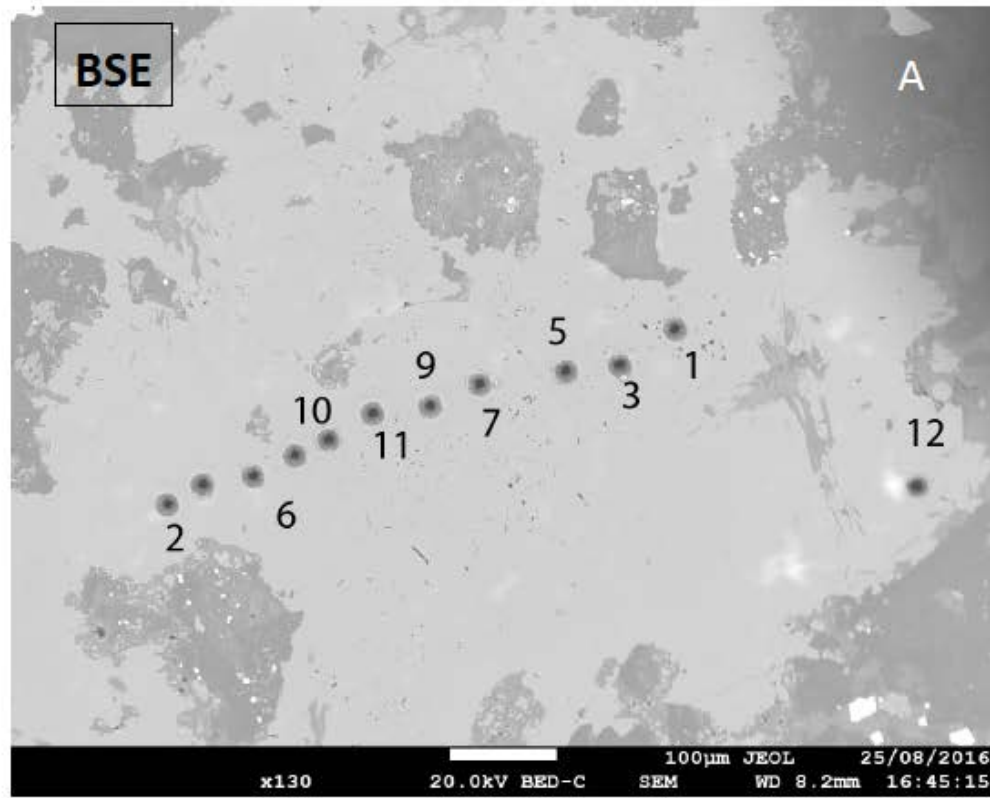


- A = Ca^{2+} , Sr^{2+} , Pb^{2+} , Ba^{2+} , Mg^{2+} , Fe^{2+} , **REE³⁺**, Eu^{2+} , Cd^{2+} , Na^+ , Al^{3+} , Y^{3+} , K^+ , Mn^{2+} , Cu^{2+}
- X = P^{5+} , Si^{4+} , **S^{4/6+}**, As^{5+} , V^{5+} , C^{4+}
- Z = **F⁻**, **Cl⁻**, **S^{2-/1-}**, OH^-

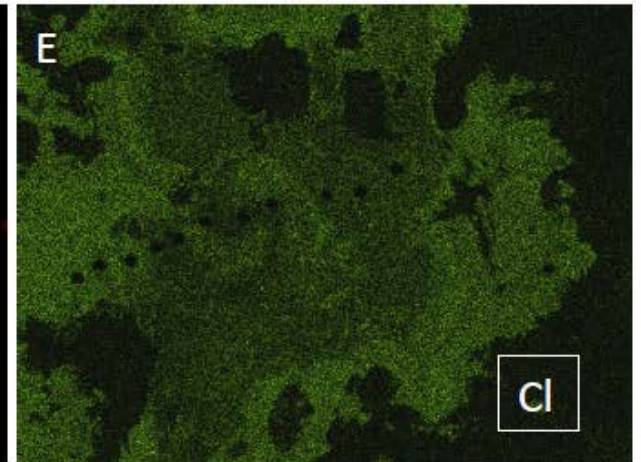
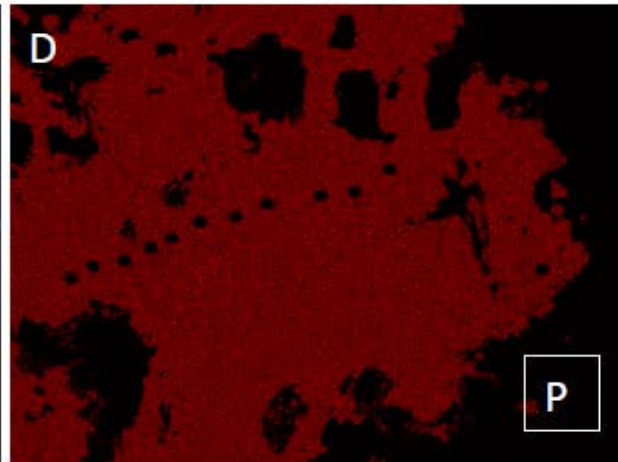
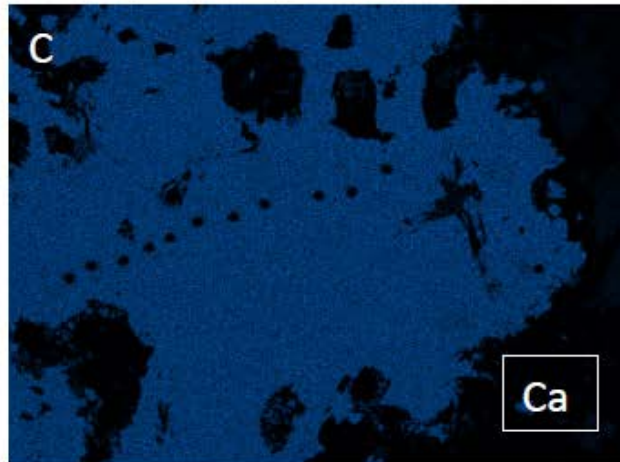
- Major (F, Cl, OH) and trace element chemistry can provide insights into fluid history.



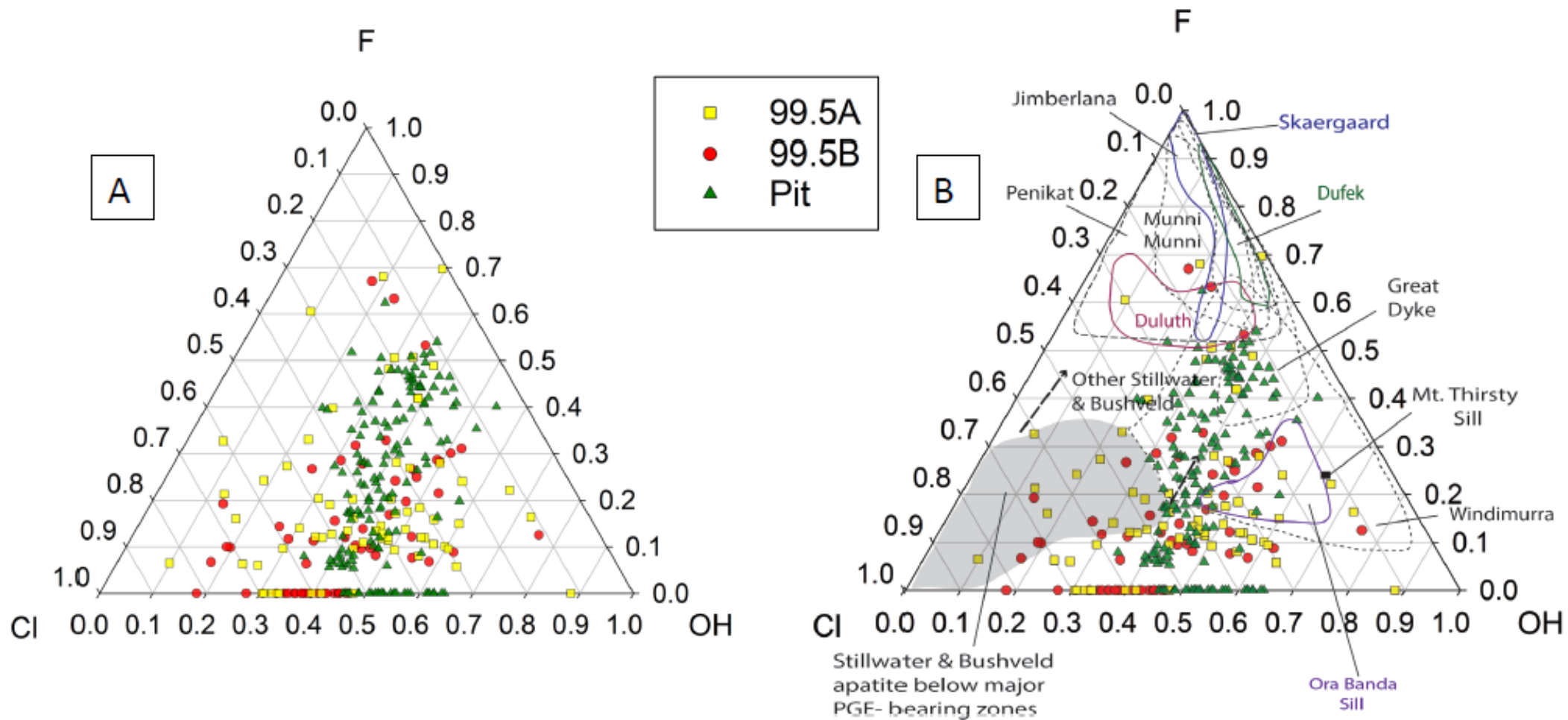
Pit sample

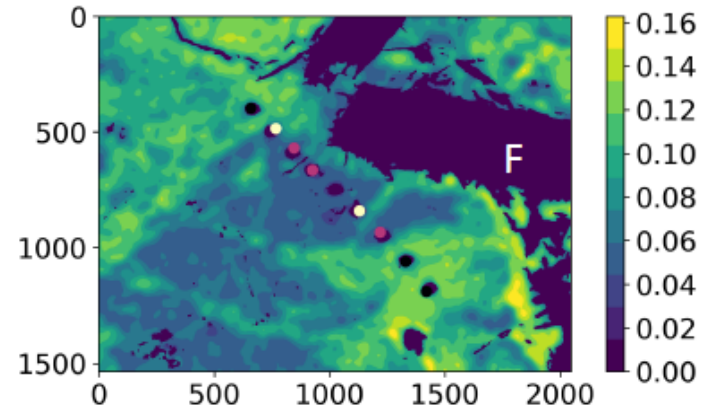
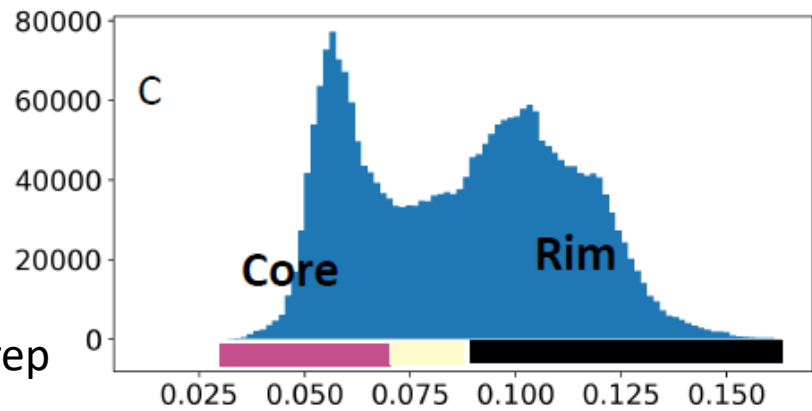
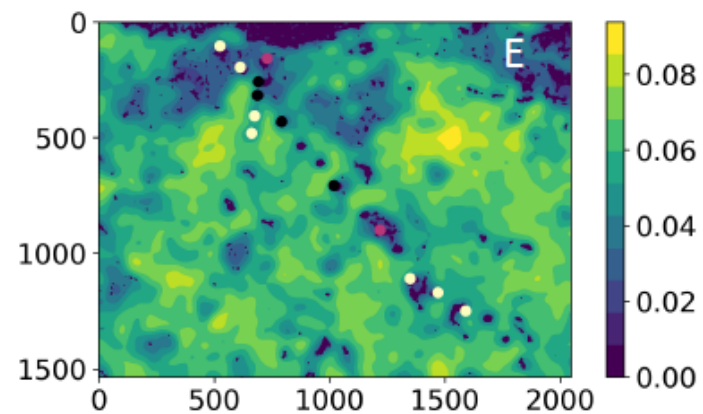
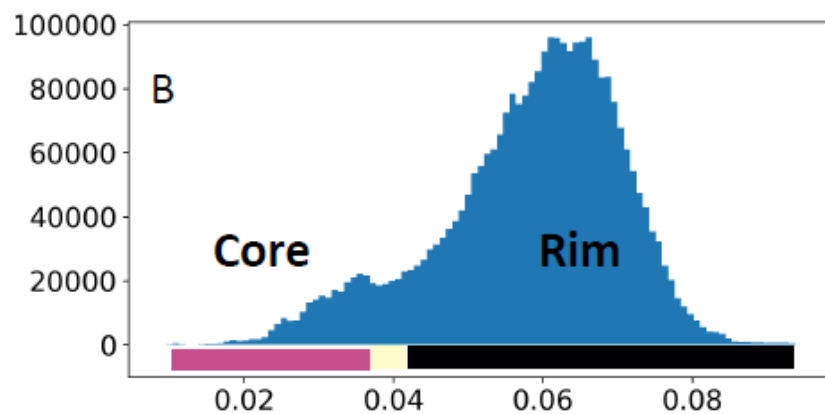
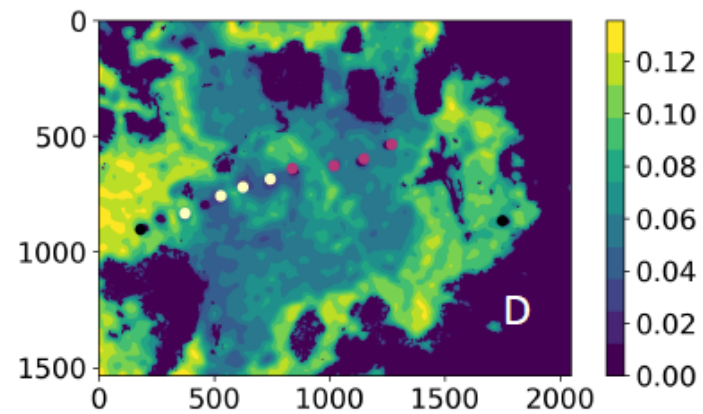
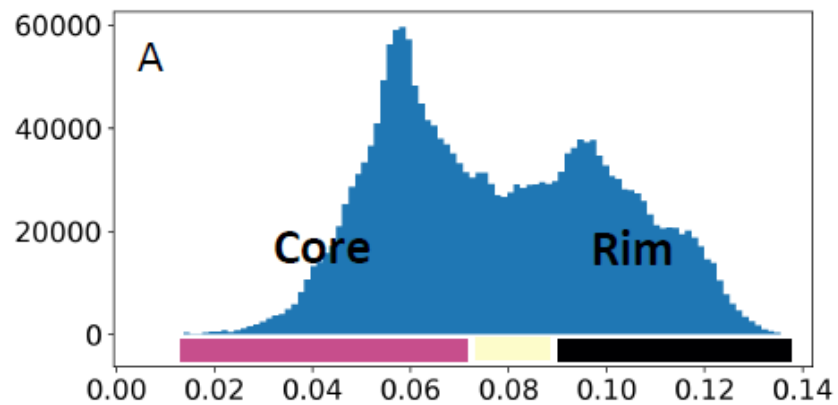


EDS Element Maps

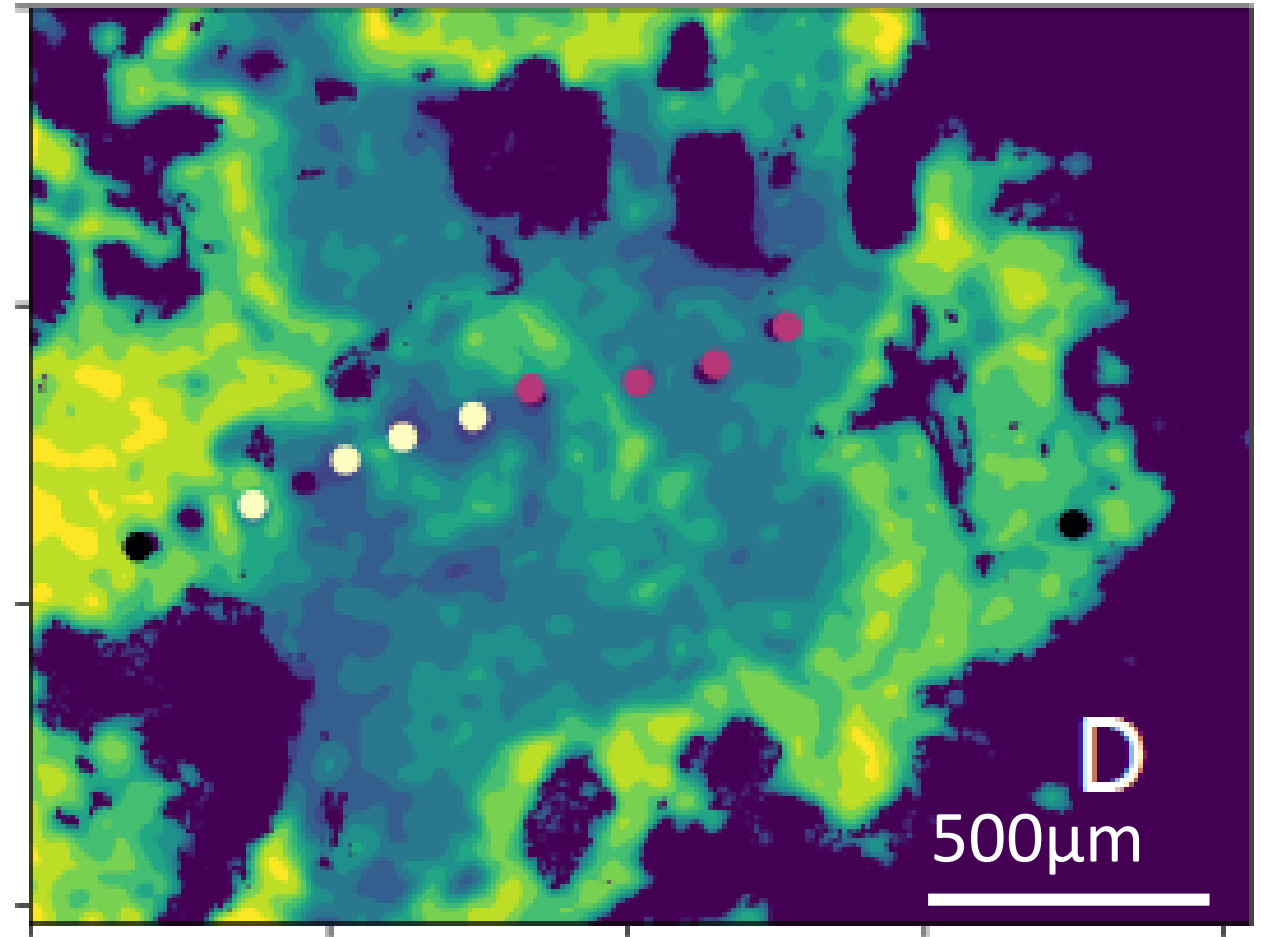
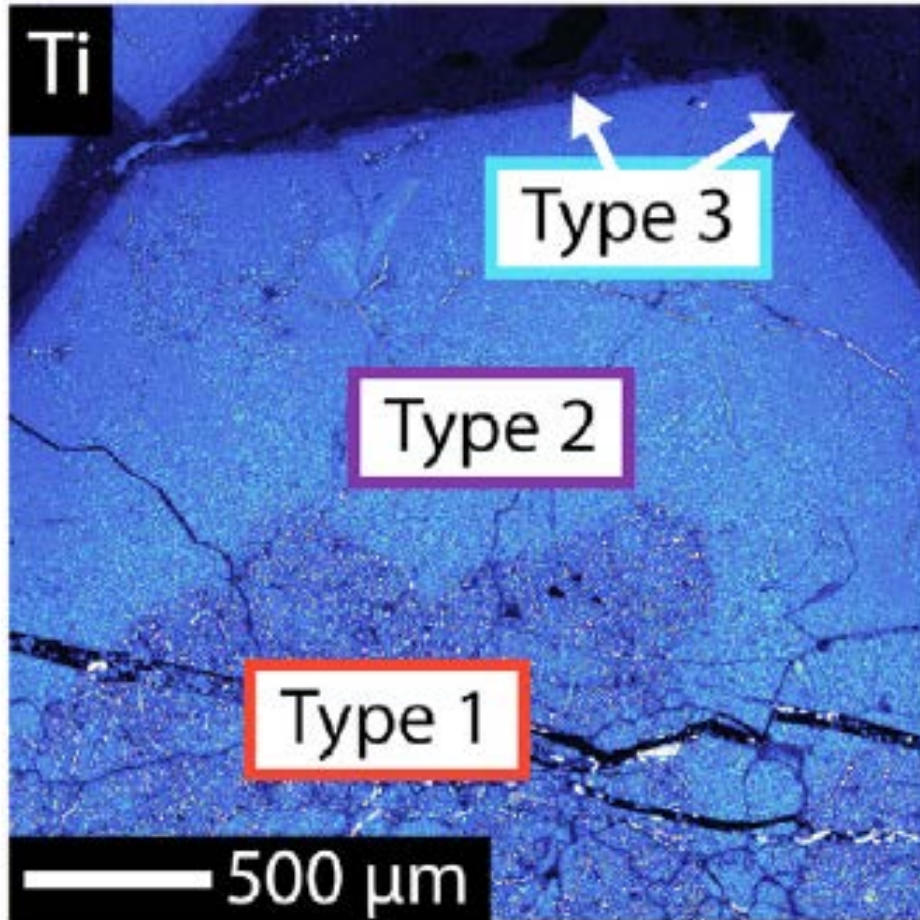


La Cruz et al. in prep





Similarities with magnetite and apatite chem



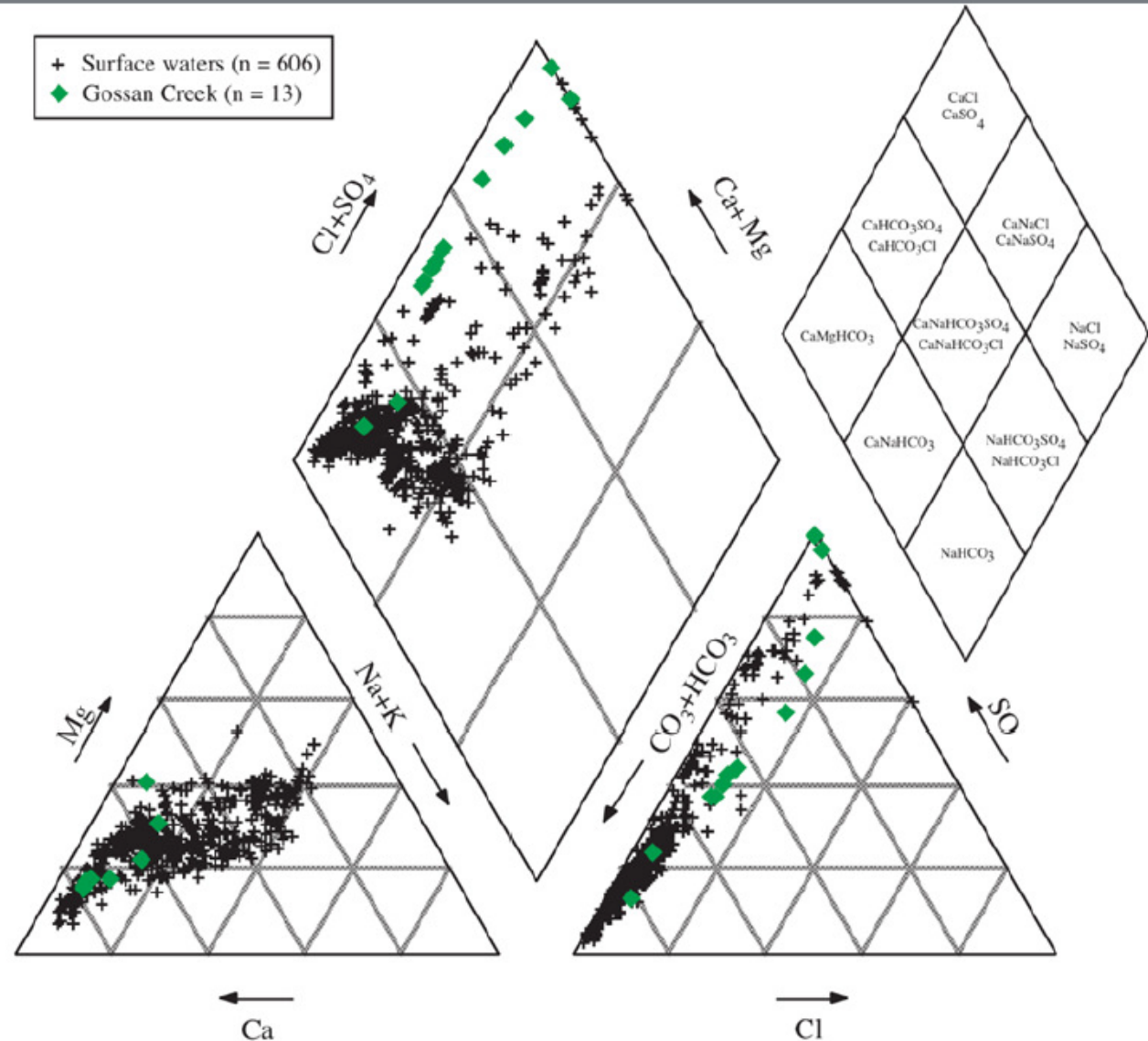
Aqueous geochem for exploration

Water chemistry

Major ions in water – typically not useful for direct exploration, but can help identify changes in water sources, water-rock reactions.

Here > 600 surface water samples – a subset has anomalously elevated SO_4 – and elevated Au

Leybourne 2018



Biogeochemistry as a tool for exploration

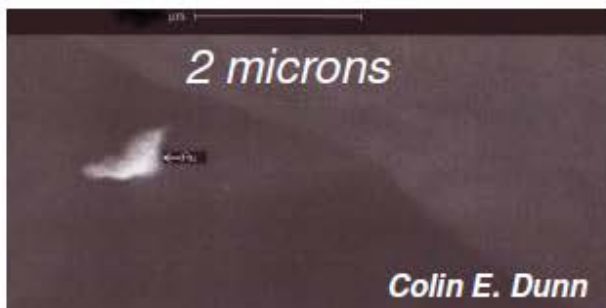
Base Metals Southern British Columbia



Dunn 2018

© Colin E. Dunn

Metal Uptake and Transport in Trees



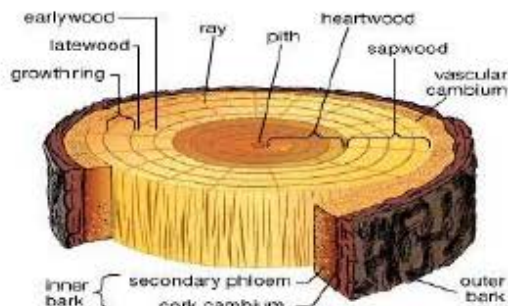
Gold in outer bark

Transport in vascular system

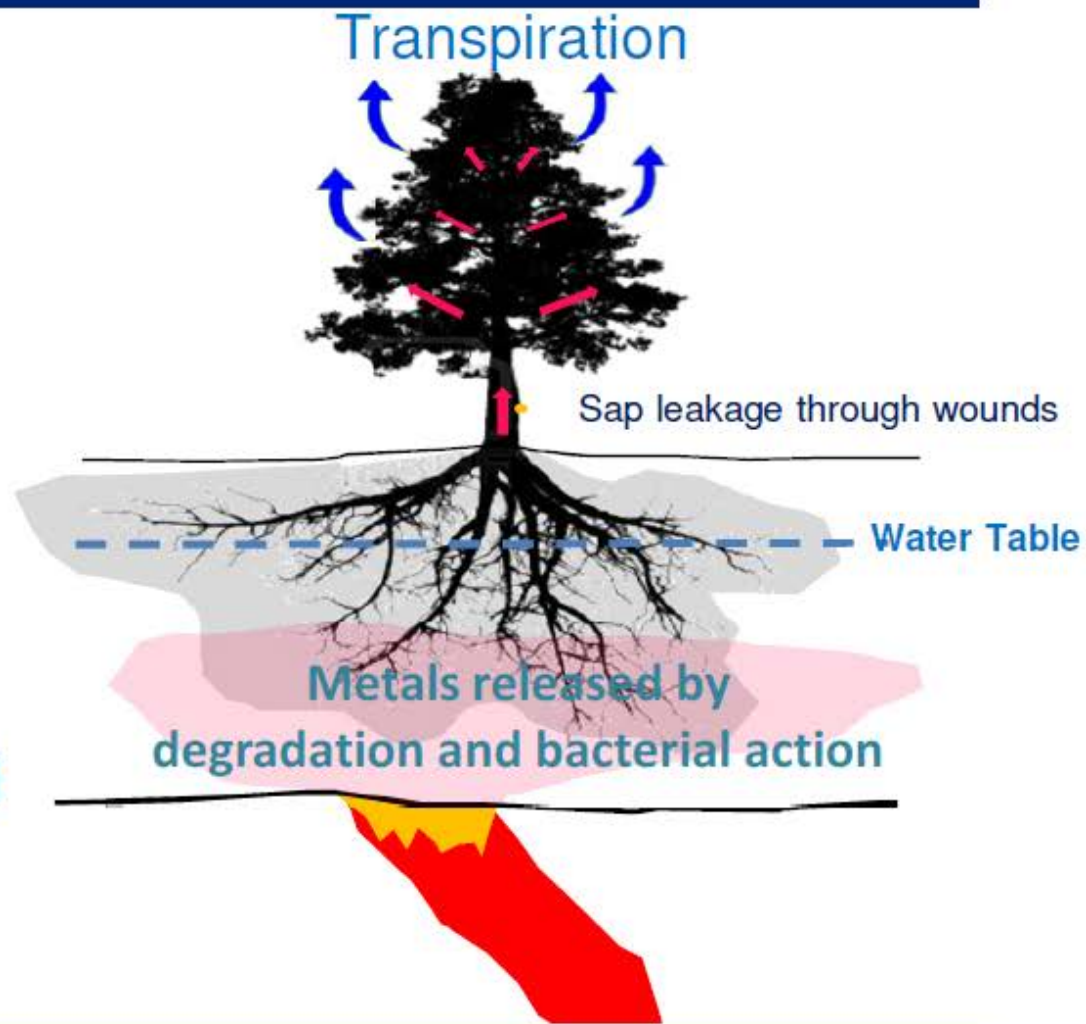


Colin E. Dunn

Incorporation into Tissues



Metal uptake facilitated by mycorrhizae



Douglas-fir Treetop Sampling



Sample Collection

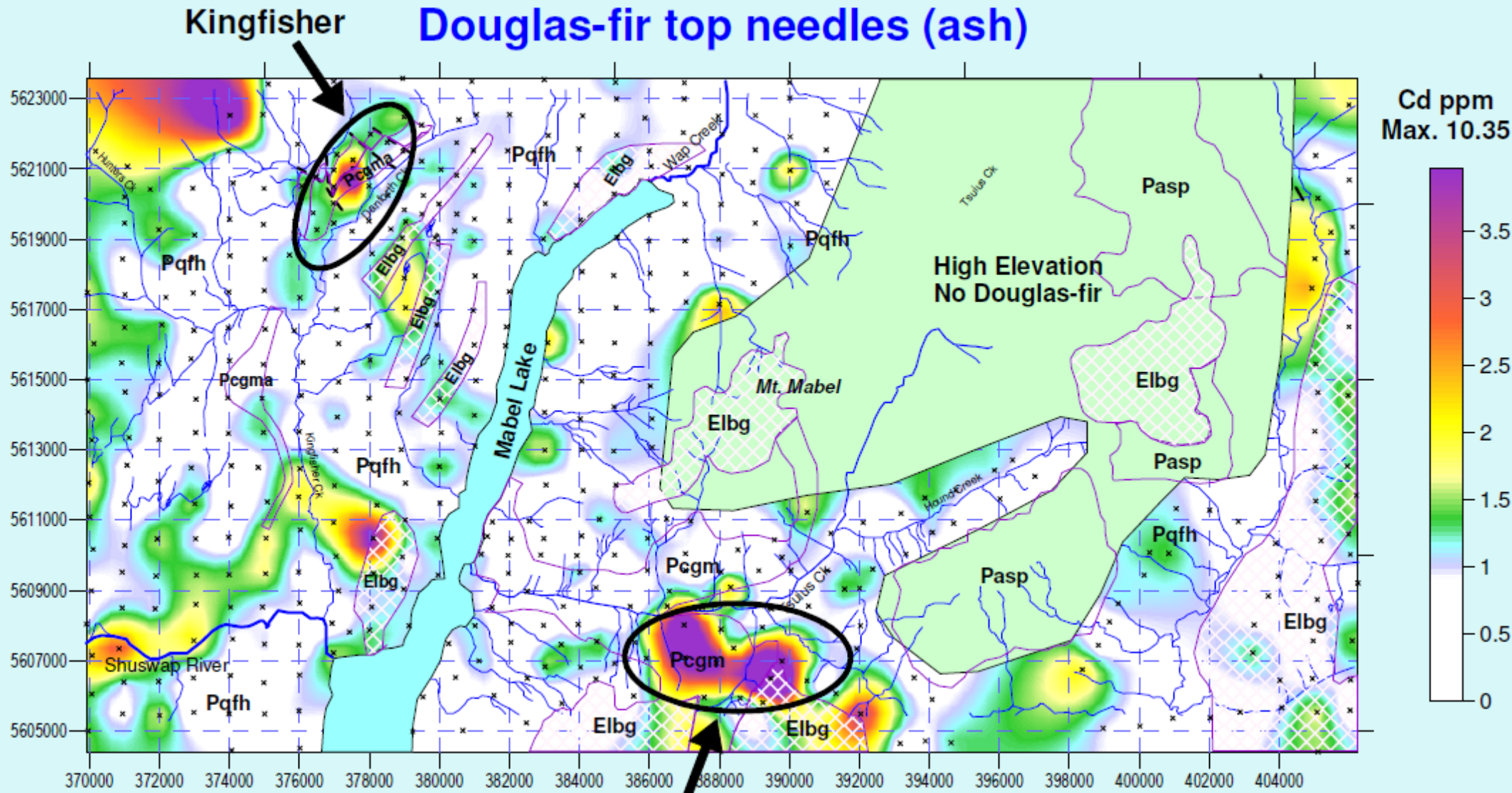


Separation of Needles from Twigs



CADMIUM

Douglas-fir top needles (ash)



Dunn 2018

South Tsuius Anomaly

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Bark Sampling

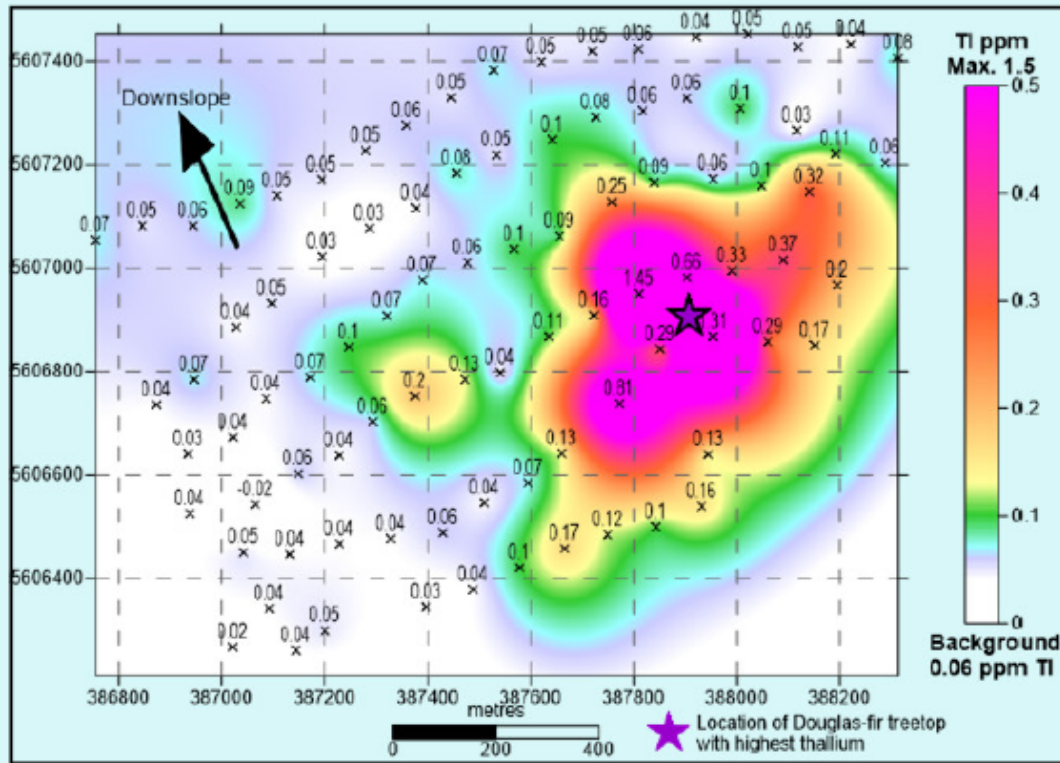


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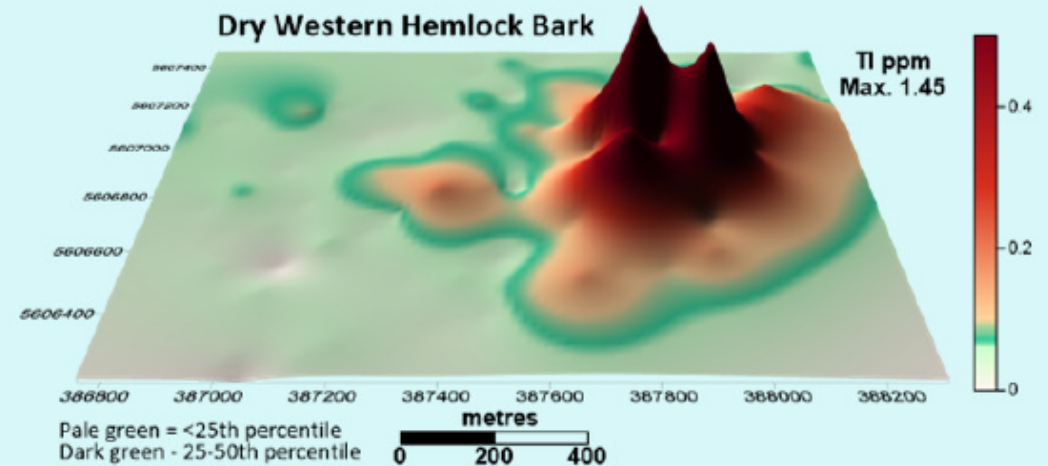


THALLIUM – Dry Western Hemlock Bark South of Tsuius Ck

Contoured percentiles

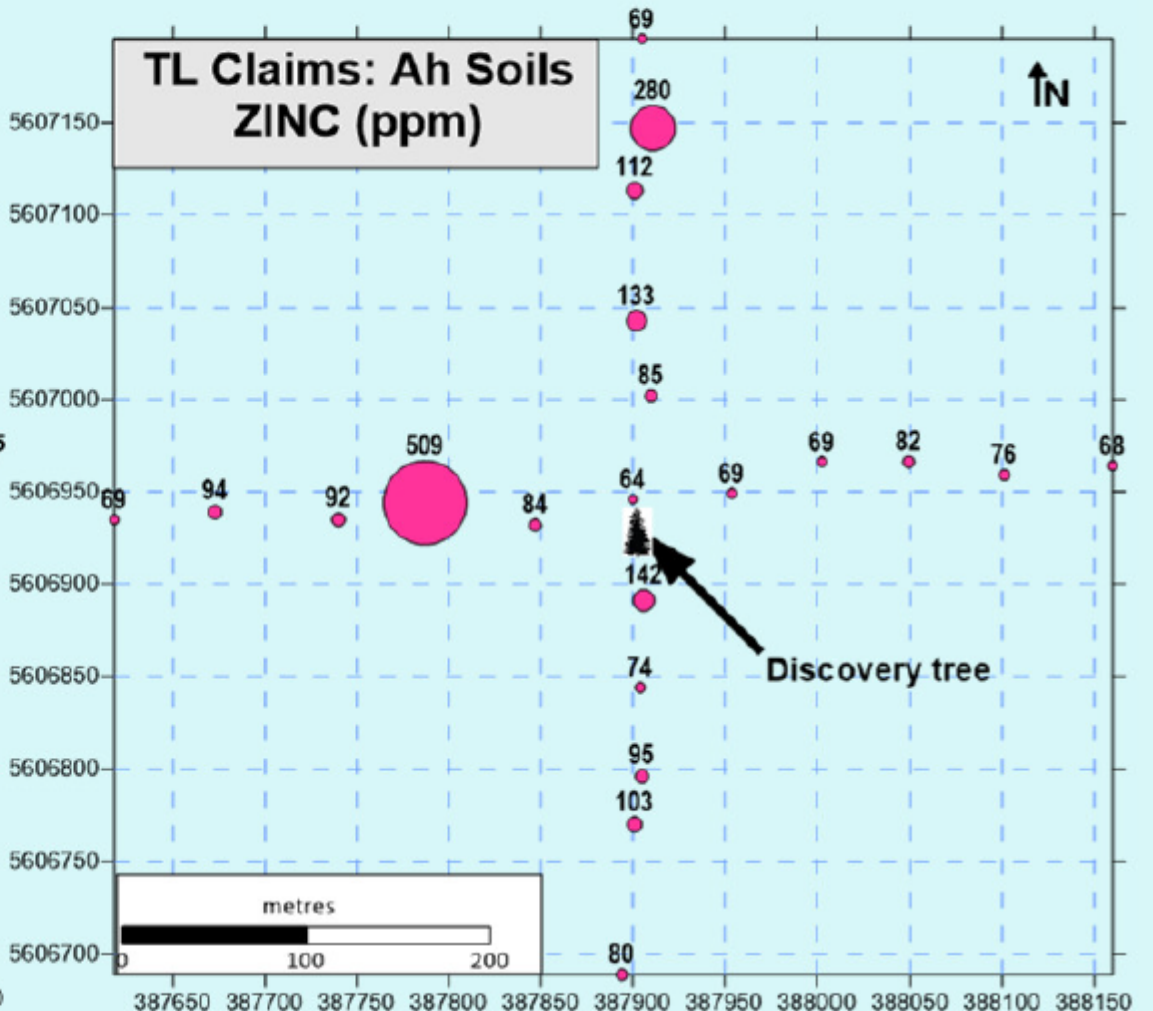
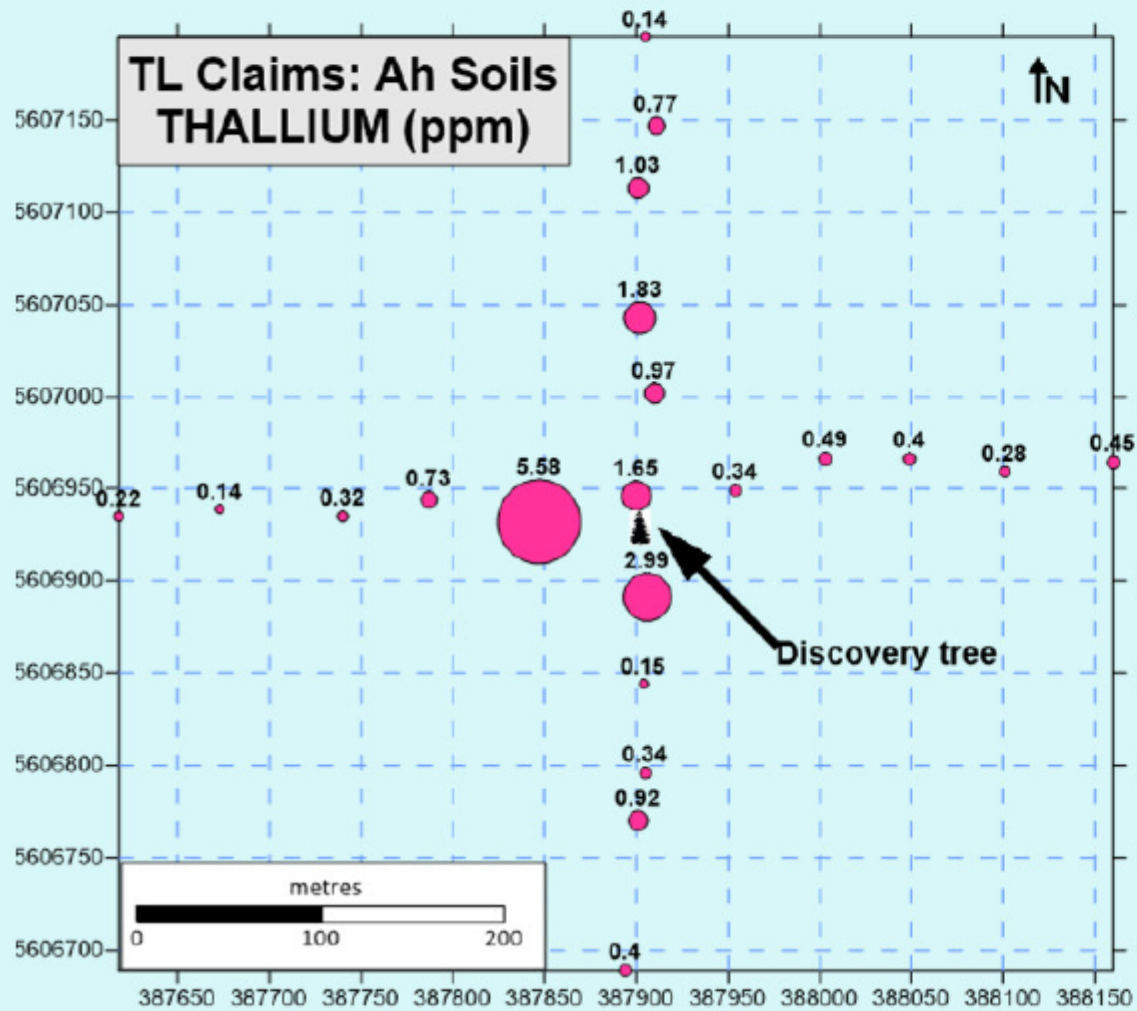


3D Surface Image

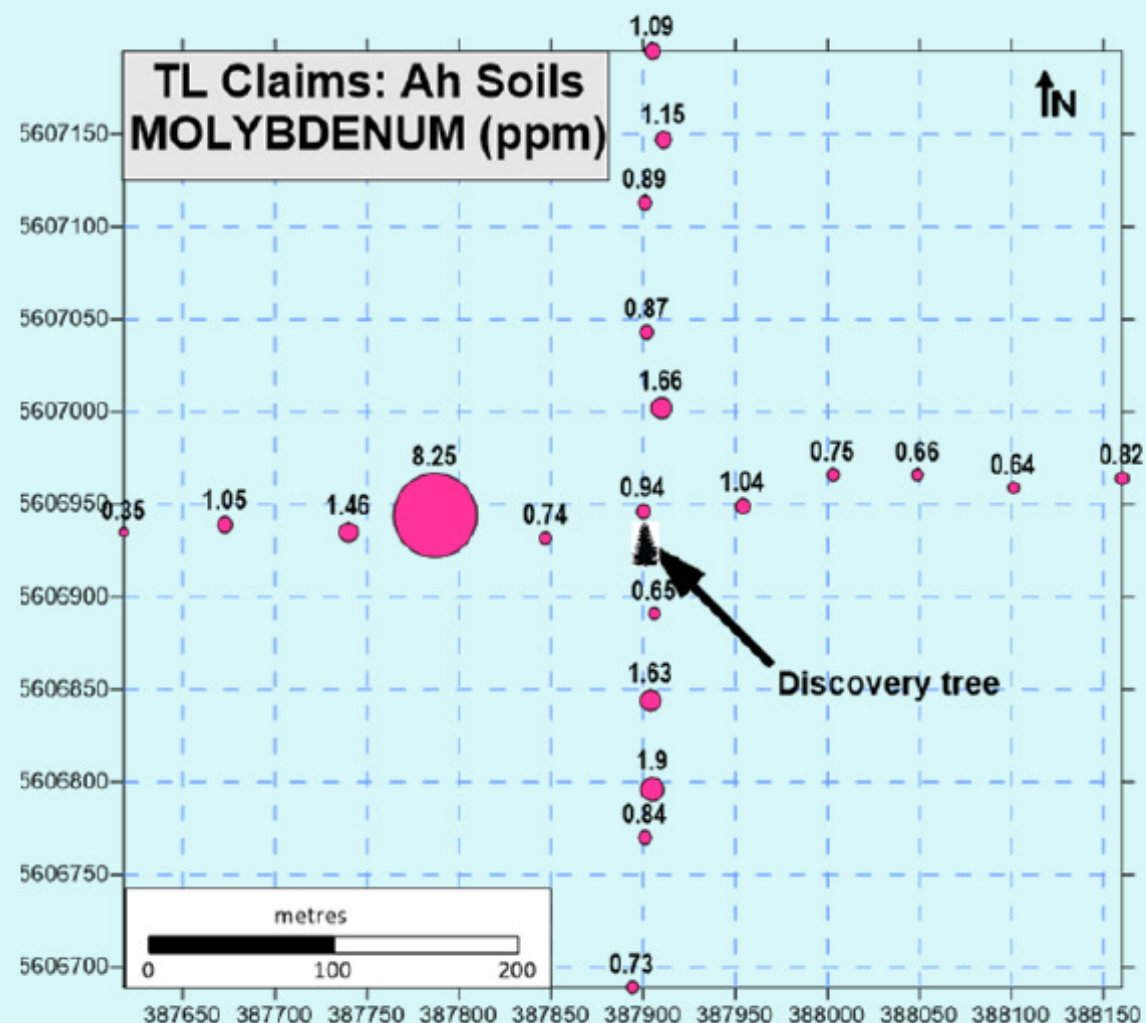
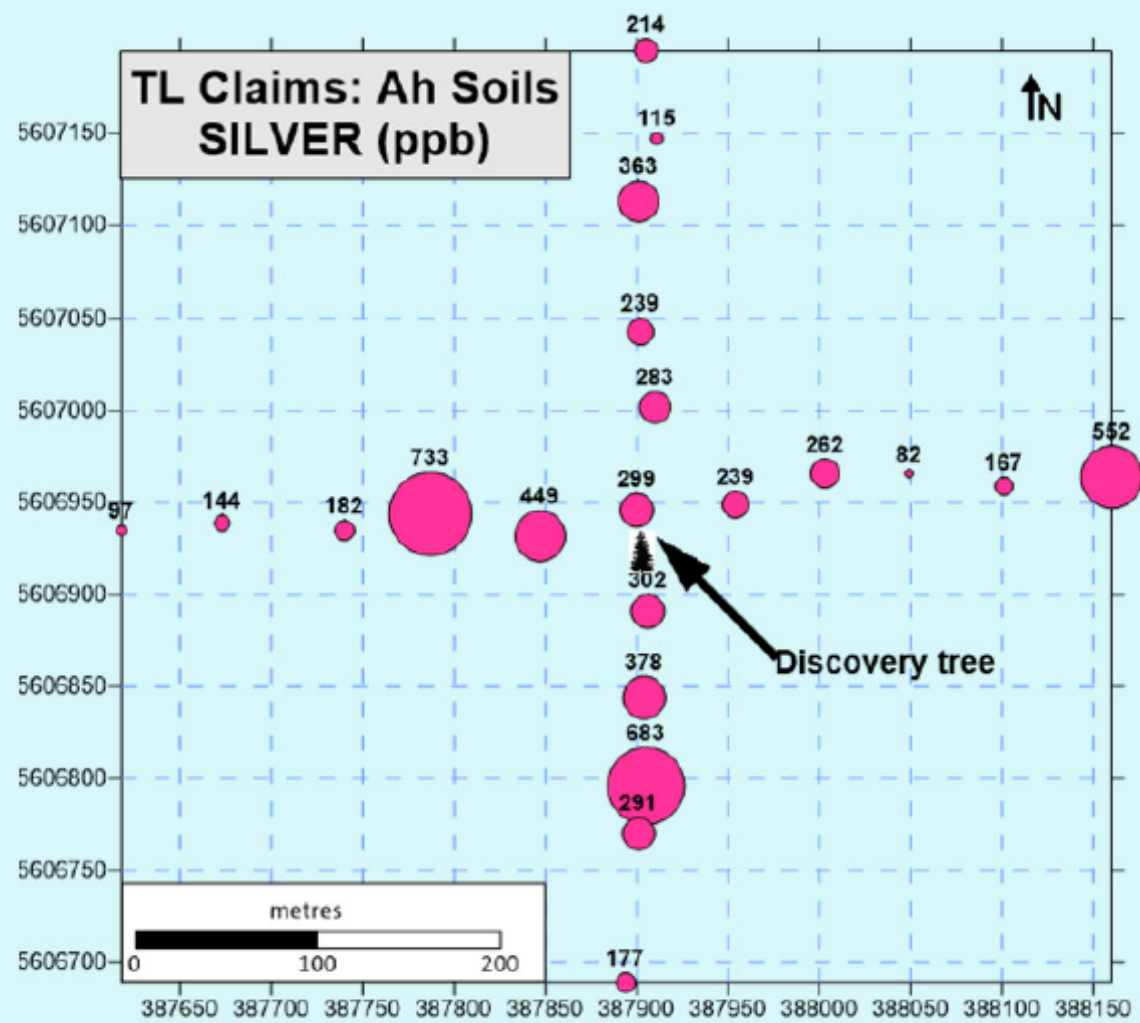


Continuous rock trench – 3 trenches





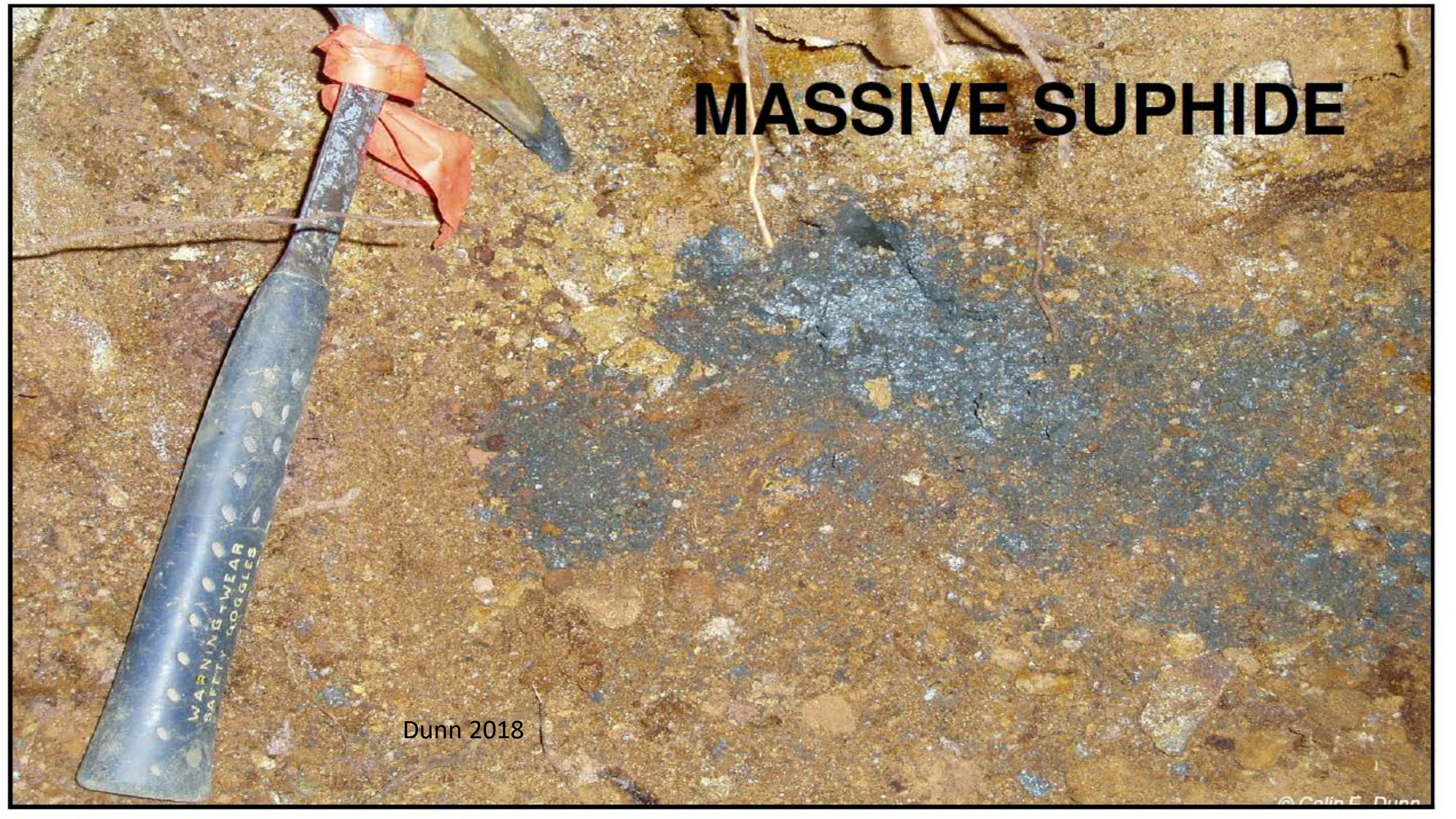
Dunn 2018



Dunn 2018

MASSIVE SUPHIDE

Dunn 2018



Summary of metal enrichments

- **Zinc** - best continuous section = 3m @ 8.98%
- **Rhenium** (580ppb)
- **Molybdenum** (max 1339 ppm)
- **Anomalous Cu, Pb, Bi, Ni, V, (Sn, W)**

A Diamond Mine Udachnaya Pipe, an open-pit diamond mine in Russia, is more than 600 meters (1,970 ft) deep.



Arakaka trend, northwestern Guyana



Medium scale mining in northwestern Guyana



Medium scale mining in northwestern Guyana



Summary

- Ore deposits and mining are important for society to exist.
- If we don't grow it, we must mine it. (We also mine to be able to grow things.)
- Can we figure out better ways to mine and regulate mining in order to mitigate harmful effects?
- The resources we need come from different types of deposits that exist globally.
- Geochemistry is a useful tool for finding these deposits.